

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

# CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

## Matter—Properties and Changes

### Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
---------------------	--------------------	-----------------------

Matter is anything with (1) mass and volume. A (2) substance is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) properties that can be observed. Substances have both physical and chemical properties. (4) Physical properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) density are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) chemical properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either physical or chemical.

- |                 |   |
|-----------------|---|
| <u>physical</u> | 7. Chemical formula $H_2O$                            |
| <u>chemical</u> | 8. Irons react vigorously when exposed to moist air   |
| <u>chemical</u> | 9. Remains unchanged when in the presence of nitrogen |
| <u>physical</u> | 10. Colorless   |
| <u>physical</u> | 11. Solid at normal temperatures and pressures        |
| <u>chemical</u> | 12. Ability to combine with another substance         |
| <u>physical</u> | 13. Melting point                                     |
| <u>physical</u> | 14. Liquid at normal temperatures and pressures       |
| <u>physical</u> | 15. Boiling point is $100^\circ C$                    |
| <u>physical</u> | 16. Conducts electricity                              |
| <u>physical</u> | 17. Density is $\frac{1}{2} \frac{g}{cm^3}$           |

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# CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

### Section 3.1 continued

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- |              |  |
|--------------|--|
| <u>true</u>  | 21. All matter that we encounter in everyday life exists in one of these physical forms. |
| <u>true</u>  | 22. A solid has definite shape and volume.   |
| <u>false</u> | 23. A liquid has a definite shape and takes on the volume of its container.              |
| <u>true</u>  | 24. A gas has both the shape and the volume of its container.                            |
| <u>false</u> | 25. The particles in a gas cannot be compressed into a smaller volume.                   |
| <u>false</u> | 26. Liquids tend to contract when heated.  |
| <u>false</u> | 27. The particles in a solid are spaced far apart.                                       |
| <u>false</u> | 28. The words gas and vapor can be used interchangeably.                                 |

# Matter And Change Continued Study Guide Answers

**B Lingard**



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