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web if you start with 14 82 g of ca oh 2 c a o h 2 and 16 35 g of h<sub>2</sub>so<sub>4</sub> h 2 s o 4 a determine the limiting reagent b determine the number of moles of h<sub>2</sub>o h 2 o produced c determine the number of grams of caso<sub>4</sub> c a s o 4 produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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