

CHEMISTRY
PAPER – 2
(PRACTICAL)

(Maximum Marks: 30)

(Time allowed: Three hours)

*(Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.)*

**ALL ANSWERS MUST BE WRITTEN IN THE ANSWER BOOKLET PROVIDED
SEPARATELY.**

*Question 1 is an oxidation-reduction titration in which sufficient working details are given.
All essential working must be shown.*

*Question 2 is an exercise dealing with identification of organic compounds.
Credit will be given for precise observations recorded and for well-drawn deductions.*

Question 3 is an exercise in qualitative analysis.

Read the questions carefully and follow the given instructions.

Attempt all questions.

*All working, including rough work, should be done on the same sheet as the
rest of the answer.*

The intended marks for questions or parts of questions are given in brackets [].

Mathematical Tables are provided.

Attempt all questions.

Question 1

[7]

You are provided with two solutions as follows:

- C-10 is a solution containing 1.95 gms of potassium manganate (VII), KMnO_4 per litre.
 - C-11 is a solution prepared by dissolving 23.2 gms of hydrated ammonium iron (II) sulphate crystals, $(\text{NH}_4)_2\text{SO}_4 \cdot \text{FeSO}_4 \cdot x\text{H}_2\text{O}$ per litre.
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