

# Atomic Structure & Isotope Worksheet

- The mass of an atom is contained mainly in its \_\_\_\_\_ and \_\_\_\_\_.
- The identity of an element is determined by its number of \_\_\_\_\_.
- Isotopes are atoms with the same number of \_\_\_\_\_ and different number of neutrons.
- The charge of an atom or ion is determined by its number of \_\_\_\_\_.
- What is the central core of an atom called?  
a) electron                      b) isotope                      c) nucleus                      d) proton
- An atom with a charge is known as an \_\_\_\_\_.  
a) isotope                      b) isomer                      c) element                      d) ion
- An atom with equal number of protons and electrons is called \_\_\_\_\_.  
a) isotope                      b) neutral atom                      c) ion                      d) nucleus
- Isotopes have \_\_\_\_\_.  
a) same number of neutrons                      b) different number of neutrons  
c) different number protons                      d) different number of electrons
- An isotope of carbon is \_\_\_\_\_.  
a) boron-12                      b) nitrogen-12                      c) carbon-14                      d) hydrogen-2
- The table below contains information about several elements. In each case, enough information has been provided for you to fill in the blanks.

Protons	Neutrons	Electrons	Element name	Is it an ION? If so, what is the charge?	Nuclear symbol
47	63	47			
15	17	15			
12	12	12			
12	13	12			
28	32	27			

# Isotopes And Ions Worksheet Answers

**Michael DiSpezio**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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