

Isotopes

The number of protons in a nucleus determines the identity of the element. For example, any atom having 6 protons will be a "carbon" atom. If we were to add an extra proton to the nucleus, we would have an entirely different element. For example,



On the other hand, if we add an extra NEUTRON to a nucleus we simply end up with the same element, just a little heavier, since the charge on the nucleus would be unchanged.

ISOTOPES of a given element have the same ATOMIC NUMBER but a *different* ATOMIC MASS.

In other words, isotopes have the same number of protons but a different number of neutrons. An isotope is identified by its mass number, the sum of the protons and neutrons. The most common isotope of Carbon has a mass number of 12 and can be written as Carbon-12, two other isotopes are Carbon-13 and Carbon-14. Despite their different mass numbers, all three carbon isotopes react the same way chemically.

PART I. Answer the questions based on the above reading.

1. What is an isotope? **Isotopes are versions of the same element. They have the same number of protons and electrons as the element but different mass numbers and number of neutrons.**
2. What does the number next to isotopes signify? **The number indicates the isotope's mass number.**
3. How can you tell isotopes of the same element apart? **They will have a different mass number and different number of neutrons.**

PART II. For each of the following isotopes, write the number of protons, neutrons, and electrons. Assume all atoms are neutral.

	Chromium-58	Chromium-63
# of protons	24	24
# of neutrons	34	39
# of electrons	24	24

	Carbon-12	Carbon-13	Carbon-14
# of protons	6	6	6
# of neutrons	6	7	8
# of electrons	6	6	6

Isotopes Pogil Answer Key

JA Banks



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Isotopes Francis William Aston, 1922

Isotope User's Guide F. E. McKinney, P. S. Baker, S. A. Reynolds, 1969

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Special Sources of Information on Isotopes U.S. Atomic Energy Commission.

Office of Isotopes Development, 1960

Table of Isotopes Richard B. Firestone, 1999-08-03 Nuclear structure and decay data for thousands of isotopes and isomers a new update of the definitive reference The 1999 Table of Isotopes booklet set features Nuclear structure and radioactive decay data for approximately 3 700 isotopes and isomers an increase of more than 100 since the 1998 Update Up to date mass chain information with more than 10% revised material The latest versions as of December 1998 of the Evaluated Nuclear Structure Data File ENSDF and Nuclear Science Reference NSR file Additional data from several evaluation sources including The Table of Superdeformed Nuclear Bands and Fission Isomers Updated isotope summary table as well as energy ordered gamma ray and alpha particle tables now included on the software available from the book's website at www.wiley-vch.de/books/info/0_471_35633_6 Updated appendices for elemental data nuclear charts and gamma ray energy standards Adoption of the 1997 IUPAC recommended heavy element names More than 25 000 references Convenient links to additional atomic mass nuclear astrophysics rates spontaneous fission thermal neutron capture and more Plus the Isotope Explorer 2.22 software lets you search the entire database by level scheme drawings annotated tables data plots nuclear structure charts and keywords as well as download the latest data directly from the Table of Isotopes Web site The 1999 Update booklet features a new comprehensive isotope nuclear structure table In addition it provides clear step by step instructions on navigating the Table of Isotopes electronic content and accessing its Web site System Requirements PC Macintosh or UNIX systems with double speed and sufficient RAM to run Adobe TM Acrobat see Adobe Acrobat Reader information in the book for specific system requirements Included Adobe Acrobat Reader 3.02 for Windows 95 98 and NT and Linux 1.2.13 or higher Acrobat Reader 3.01 for Windows 3.1 Isotope Explorer 2.22 for Windows 95 98 and NT **Table of Isotopes** Richard B. Firestone, 1996 An Introduction to Isotopes and Radiations P. A. Wahid, 2001 Isotopes F. W. Aston, 1965 **Handbook of Stable Isotope Analytical Techniques** Pier A. de Groot, 2004-10-27 Parent with price Volume I contains subjective reviews specialized and novel technique descriptions by guest authors Part 1 includes contributions on purely analytical techniques and Part 2 includes matters such as development of mass spectrometers stability of ion sources standards and calibration correction procedures and experimental methods to obtain isotopic fractionation factors Volume II will be available in 2005 **Isotopes Information Center Keyword Thesaurus** United States. Isotopes Information Center, 1970 *Table of Isotopes, 2 Volume Set* Richard B. Firestone, 1997-11-28 Available to registered users of the original 2 volume set and also bundled with new copies of this publication this Update combines a booklet and CD ROM boasting more than 100 additional isotopes as well as updated appendices that include elemental data nuclear charts and gamma ray energy standards data Improved Isotope Explorer

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Inventory of Stable Isotopes C. P. Keim, H. W. Savage, B. S. Weaver, 1949 **Handbook of Stable Isotope Analytical Techniques Vol II** Pier A. de Groot, 2008-11-10 This two volume reference serves as a handbook containing a wealth of information for all isotope chemists working in a wide range of disciplines including anthropology to ecology drug detection methodology to toxicology nutrition to food science and the atmospheric sciences to geochemistry Complementing the first volume Volume II includes matters that are not strictly confined to the analytical techniques themselves but relate to analysis of stable isotopes such as the views on the development of mass spectrometers isotopic scales standards and references and directives for setting up a laboratory ALSO AVAILABLE Volume I Dec 2004 0444511148 9780444511140 176 00 Volume I and II set Oct 2007 0444511164 9780444511164 205 00 Presents an encyclopedic overview of stable isotope analytical techniques in an objective way Includes descriptions of methods and diagrams of analytical devices Addresses how older techniques formed the basis for present day techniques which can be useful in constructing modern analytical systems Complements Volume I of the set *Special Sources of Information on Isotopes* U.S. Atomic Energy Commission. Division

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