

# Mark scheme

## Chemistry Past Paper Questions 1

### SECTION B: CHEMISTRY

Item	Expected answers	Additional guidance	Marks
4 (a)	5		1
4 (b)	An electron has a much smaller mass than a proton.		1
4 (c)	$(20 \times 10) + (80 \times 11)/100$ (1) = 10.8 (1)	$(200 + 80)/100$ (1) or $1080/100$ (1)	2
4 (d)	2 or 8 but not 3 for 1 mark. 2, 8, 3 for 2 marks.		2
<b>Total mark</b>			<b>6</b>

Item	Expected answers	Additional guidance	Marks
5 (a)	Any <b>one</b> of the following: It is on the right hand side of the periodic table (1) It is an inert gas/rare gas/noble gas (1)		1
5 (b)	Any <b>two</b> of the following: They have similar reactions to each other (1) They have similar properties to each other (1) They all have 1 electron in their outer shell (1)	Do not accept 'they are all alkali metals'	2
<b>Total mark</b>			<b>3</b>

# June 2013 Chemistry Paper 1 Mark Scheme

**N Noddings**



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1. Understanding the eBook June 2013 Chemistry Paper 1 Mark Scheme
  - The Rise of Digital Reading June 2013 Chemistry Paper 1 Mark Scheme
  - Advantages of eBooks Over Traditional Books
2. Identifying June 2013 Chemistry Paper 1 Mark Scheme
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an June 2013 Chemistry Paper 1 Mark Scheme
  - User-Friendly Interface
4. Exploring eBook Recommendations from June 2013 Chemistry Paper 1 Mark Scheme
  - Personalized Recommendations
  - June 2013 Chemistry Paper 1 Mark Scheme User Reviews and Ratings
  - June 2013 Chemistry Paper 1 Mark Scheme and Bestseller Lists
5. Accessing June 2013 Chemistry Paper 1 Mark Scheme Free and Paid eBooks
  - June 2013 Chemistry Paper 1 Mark Scheme Public Domain eBooks
  - June 2013 Chemistry Paper 1 Mark Scheme eBook Subscription Services
  - June 2013 Chemistry Paper 1 Mark Scheme Budget-Friendly Options
6. Navigating June 2013 Chemistry Paper 1 Mark Scheme eBook Formats

- ePub, PDF, MOBI, and More
  - June 2013 Chemistry Paper 1 Mark Scheme Compatibility with Devices
  - June 2013 Chemistry Paper 1 Mark Scheme Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of June 2013 Chemistry Paper 1 Mark Scheme
  - Highlighting and Note-Taking June 2013 Chemistry Paper 1 Mark Scheme
  - Interactive Elements June 2013 Chemistry Paper 1 Mark Scheme
8. Staying Engaged with June 2013 Chemistry Paper 1 Mark Scheme
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers June 2013 Chemistry Paper 1 Mark Scheme
9. Balancing eBooks and Physical Books June 2013 Chemistry Paper 1 Mark Scheme
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection June 2013 Chemistry Paper 1 Mark Scheme
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine June 2013 Chemistry Paper 1 Mark Scheme
- Setting Reading Goals June 2013 Chemistry Paper 1 Mark Scheme
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of June 2013 Chemistry Paper 1 Mark Scheme
- Fact-Checking eBook Content of June 2013 Chemistry Paper 1 Mark Scheme
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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