

$$\text{percent yield} = \frac{\text{actual yield (g)}}{\text{theoretical yield (g)}} \times 100\%$$

↳ from B-E-A chart

Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation.
- Identify the given (with units) and what you want to find (with units).
- Show set up with units. Give final answer with units and label.

1. Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



1 mol H_2O : 18.02 g/mol

36 g $\frac{1\text{ mol H}_2\text{O}}{18.02\text{ g}} = 1.99\text{ mol H}_2\text{O}$

1 mol H_2 : 2.02 g

2 mol $\text{H}_2 = \boxed{4.04\text{ g H}_2}$

2. Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



1 mol $\text{NaCl} = 58.44\text{ g}$
 $581\text{ g NaCl} \times \frac{1\text{ mol NaCl}}{58.44\text{ g}} = 9.94\text{ mol NaCl}$

1 mol $\text{Na} = 22.99\text{ g}$

$9.94\text{ mol Na} \times 22.99\text{ g} = \boxed{228.52\text{ g Na}}$

Modeling Chemistry Unit 12

Cornell University



Modeling Chemistry Unit 12:

ACH, Models in Chemistry ,1928 **Resources in Education** ,1984-10 **Science Software** ,1989 *Energy Research Abstracts* ,1987 Selected Solutions for Chemistry, Concepts and Models by Robinson, Odom, and Holtzclaw John H. Meiser,1992 *Energy Abstracts for Policy Analysis* ,1986 *The World of Chemistry* Melvin D. Joesten,David O Johnston,John T Netterville,1988 SUMMARY Consists of 26 programs on various chemistry topics including the atom chemical bonds the periodic table water etc Models in Chemical Science George Simms Hammond,1971 **Register ...** California. University,University of California, Berkeley,1909 **Commerce Business Daily** ,2001-05-22

Comprehensive Medicinal Chemistry Corwin Hansch,Peter George Sammes,John Bodenhan Taylor,1990 Drug design is a multi disciplinary activity involving chemists biologists biochemists pharmacologists and many others the chemist s role is central in inventing new compounds which exert a beneficial effect However once a lead for a new active drug has been established its effective delivery has to be demonstrated and extensive toxicological studies undertaken to demonstrate its safety before clinical trials can commence The metabolic fate of the drug has to be revealed and detailed distribution studies carried out in order to satisfy the regulatory authorities before the new compound can be marketed Comprehensive Medicinal Chemistry describes all these aspects of the design of a drug whilst centering on the chemical mechanism whereby such agents act Volume 4 covers quantitative drug design **Register** University of California, Berkeley,1917 **Circular of Information with Reference Primarily to the Undergraduate Division at Berkeley** University of California, Berkeley,1917 *Register of the University of California* University of California, Berkeley,1909 The City Record New York (N.Y.),1916 Cornell University Announcements Cornell University,1921 *Catalogue ...* Mills College,1919

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This book delves into Modeling Chemistry Unit 12. Modeling Chemistry Unit 12 is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Modeling Chemistry Unit 12, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Unit 12
- Chapter 2: Essential Elements of Modeling Chemistry Unit 12
- Chapter 3: Modeling Chemistry Unit 12 in Everyday Life
- Chapter 4: Modeling Chemistry Unit 12 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Modeling Chemistry Unit 12. The first chapter will explore what Modeling Chemistry Unit 12 is, why Modeling Chemistry Unit 12 is vital, and how to effectively learn about Modeling Chemistry Unit 12.

3. In chapter 2, the author will delve into the foundational concepts of Modeling Chemistry Unit 12. This chapter will elucidate the essential principles that must be understood to grasp Modeling Chemistry Unit 12 in its entirety.

4. In chapter 3, the author will examine the practical applications of Modeling Chemistry Unit 12 in daily life. The third chapter will showcase real-world examples of how Modeling Chemistry Unit 12 can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Modeling Chemistry Unit 12 in specific contexts. This chapter will explore how Modeling Chemistry Unit 12 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Unit 12. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Modeling Chemistry Unit 12.

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