

12.3 Limiting Reagent and Percent Yield



BY:
MICHELLE ALICE GANIAN

Reaction Stoichiometry

Amount of Product Formed and Percent Yield



Theoretical Yield:	0.074 g O ₂	
Actual Yield:	0.069 g O ₂	
Percent Yield:	$\frac{0.069 \text{ g O}_2}{0.074 \text{ g O}_2}$	= 93.2% Yield

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Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **AP Chemistry Crash Course, Book + Online** Adrian

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writing this Crash Course Mr Dingle has written The Periodic Table Elements With Style How To Make A Universe With 92 Ingredients and SAT Chemistry Crash Course He is the 2011 winner of the School Library Association of the UK s Information Book Award and in 2012 was honored with the prestigious literary prize Wissenschaftsbuch des Jahre sponsored by the Austrian Ministry of Science and Research

Concepts And Problems In Physical Chemistry P.S. Raghavan,1997
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Study Guide [to] Chemistry, a Systematic Approach [by] Sisler, Dresdner, and Mooney William H. Myers,William T. Mooney,1980

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