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$$m_{mix} = \frac{m_1 + m_2}{V_1 + V_2} = \frac{1.5 \times 480 + 1.2 \times 520}{480 + 520} = \frac{720 + 624}{1000} = 1.344 \text{ m}$$
 q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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