



Metric Conversion Worksheet Answers

*mm and cm, cm and m, m and km
up to 10 with up to 1 decimal place*

Created by the Math Salamanders www.math-salamanders.com

1) 7000 m = **7** km

2) 1 km = **1000** m

3) 2.4 km = **2400** m

4) **820** cm = 8.2 m

5) 3.1 km = **3100** m

6) 810 cm = **8.1** m

7) 700 cm = **7** m

8) **54** mm = 5.4 cm

9) 7.7 cm = **77** mm

10) **28** mm = 2.8 cm

11) **10** km = 10000 m

12) **3.1** km = 3100 m

13) 4.2 cm = **42** mm

14) **8.8** m = 880 cm

15) **9.5** cm = 95 mm

16) **10300** m = 10.3 km

17) 6.7 km = **6700** m

18) 10.1 m = **1010** cm

19) 7.3 cm = **73** mm

20) **9** m = 900 cm

21) 4.3 m = **430** cm

22) 2 cm = **20** mm

23) **8.7** km = 8700 m

24) 500 cm = **5** m

25) **6.1** m = 610 cm

26) **90** mm = 9 cm

27) **99** mm = 9.9 cm

28) 2 cm = **20** mm

29) **580** cm = 5.8 m

30) **8.2** m = 820 cm

Metric Conversion Practice 2 Answer

Joyce LeFever Kee, Sally M. Marshall

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

Metric Conversion Practice 2 Answer:

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integrated forensic theme updated features and an expanded media supplements package **Lab Manual for General, Organic, and Biochemistry** Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students **Basic Drug Calculations** Meta Brown, Meta Brown Seltzer, 1979 **Skills for Professional Nursing Practice** Barbara A. Norton, Anna Miller, 1986 **Clinical Calculations** Joyce LeFever Kee, Sally M. Marshall, 1996 This student friendly text helps you to brush up on your math skills and learn to perform drug calculations with confidence After guiding you step by step through the systems conversions and methods of dosage calculation the text will show you how to calculate dosages for oral injectable and intravenous drugs in both general and specially settings Clinical Calculations 3rd Edition makes learning dosages easy with 25% more practice

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