

Limiting and Excess Reactants

- ① For the following equation, determine which reactant is limiting and which reactant is in excess. The amount of reagents used is shown.



- ② 35.5 g of silver nitrate is reacted with 35.5 grams of sodium sulfide, producing silver sulfide and sodium nitrite.

Ⓐ Write and balance the equation.

Ⓑ Calculate the number of grams of silver sulfide produced in grams.

- ③ For the following reaction: $\text{C}_3\text{H}_8 + 5 \text{ O}_2 \longrightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$

Ⓐ Write and balance the equation.

Ⓑ How many grams of propane is this.

Limiting And Excess Reactants Worksheet Extension Questions

Lifelique

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape with a gradient, partially overlapping the bar's end.

Limiting And Excess Reactants Worksheet Extension Questions:

Limiting Reactant and Percent Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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