

Limiting Reagent Worksheet

- 1 Nitric oxide (NO) reacts with oxygen gas (O₂) to form nitrogen dioxide (NO₂) gas.



In one experiment, 0.866 mol of NO is mixed with 0.503 mol of O₂.

- Determine the limiting reagent.
- Calculate the number of moles of NO₂ produced.

- 2 Consider the following reaction:



- Suppose 0.86 mol of MnO₂ reacts with 48.2 g of HCl. Which reagent will be used first?
- How many grams of Cl₂ will be produced?

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web raw milk received according to raw milk receiving flow chart 2 page 38 of 46 penn state berkey creamery process flow 2014 frozen yogurt product name frozen yogurt plant name penn state berkey creamery ingredients milk skim milk powder sugar dry corn syrup solids stabilizer yogurt liquid flavoring ingredients packaging packaged in

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web the yoghurt samples were analyzed and compared with reconstituted full cream milk powder fcmp based yoghurt for their chemical composition microbiological and sensory qualities using standard

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web may 2 2008 in the most important step the frozen yogurt mix is sent through an ammonia cooled scraped surface heat exchanger where initial freezing takes place and air is injected to give the frozen yogurt a light creamy texture finally the frozen yogurt is packaged hardened and finally sent to a storage freezer to await shipping