Question Number	Acceptable Answers		Reject	Mark
4(f)(i)	(7.5 x 0.81) = 6.075 / 6.08 (g) Ignore sf except 1sf		6.07 Wrong units	1
Question Number	Acceptable Answers		Reject	Mark
4(f)(ii)	Look at final answer. 67% scores 3 marks; answers with 3sf rounding to 67 score 2 marks. If this is incorrect follow this scheme: METHOD 1 Mol butan-1-ol = (6.075/74) = 0.0820945 maximum mass 1-bromobutane = (0.0820945 x 137) = 11.246959 g	h (1) (1)	Percentages calculated from volumes with no conversion to mol or mass. 6.075/7.5 x 100 = 81% scores 0	3
	% yield = ((7.5/11.24659)x100 = 66.85)			
	= 67% to 2 sf	(1)		
	OR METHOD 2			
	7.5/137 = 0.0547445 mol (bromobuta	(1)	67.0 (This is 3sf)	
	6.075/74 = 0.0820945 mol butan-1-o	1		
		(1)		
	% yield = ((0.05474455)×100/0.0820945) = 66.85)			
	= 67% to 2 sf	(1)		
	Also TE from one step of the calculation to the next and TE on 4f(i) unless yield > 100%.			
	Use of 6.08 gives 0.082161 mol, 11.256216 g bromobutane, final answ 67%	er		
	11.3g bromobutane gives 66%.			

Reject

Question

Acceptable Answers

Total for Question 4 = 15 marks

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