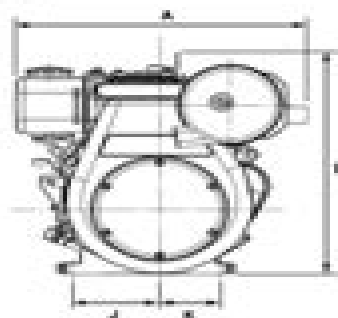
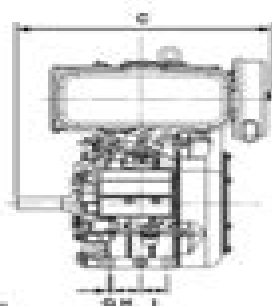
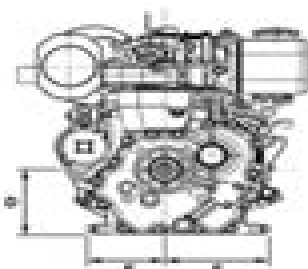


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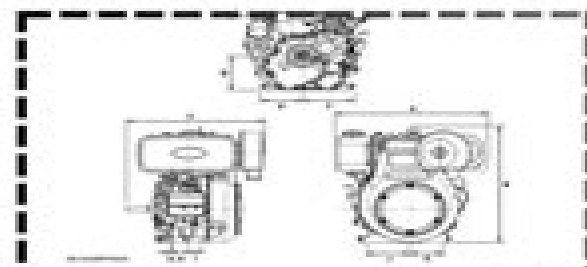
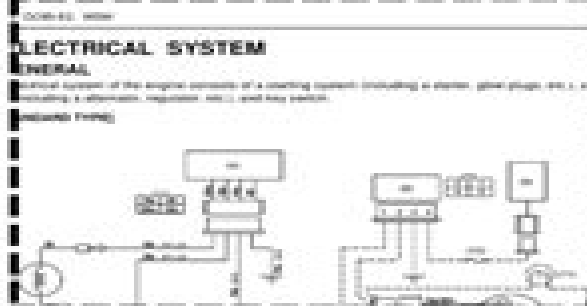
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1. Remove the oil filler cap from the top of the oil filler cap.
2. Add oil to the oil filler cap until the oil level reaches the top of the oil filler cap.
3. Tighten the oil filler cap.
4. Check the oil level after the engine has been run for 5 minutes.
5. If the oil level is low, add oil to the oil filler cap.
6. If the oil level is high, remove oil from the oil filler cap.
7. If the oil level is low, add oil to the oil filler cap.
8. If the oil level is high, remove oil from the oil filler cap.
9. If the oil level is low, add oil to the oil filler cap.
10. If the oil level is high, remove oil from the oil filler cap.



1. Remove the oil filler cap from the top of the oil filler cap.
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web feb 25 2022 detailed experimental light brown nutmeg powder 4 g was mixed with 25 ml of clear and colorless diethyl ether solvent and the brown cloudy mixture was filtered by gravity filtration with fluted fast flow filter paper in a powder funnel into a 250 ml round bottom flask rbf leaving light brown residue and beige filtrate diethyl ether was

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web trimyrustin in the major triglyceride found in nutmeg representing 20 25 of the dried weight of ground nutmeg to extract the trimyrustin we performed a solid liquid extraction using the organic solvent methylene chloride CH_2Cl_2

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web isolation of trimyrustin from nutmeg purpose the purpose of this experiment is to isolate myristic acid from ground nutmeg seeds nutmeg seeds are extracted using tH_2O to yield crude trimyrustin then recrystallized to yield pure trimyrustin which will be hydrolyzed and undergo esterification

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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyrustin is a triglyceride what are triglycerides prelab question 2 how do microwaves heat microwave procedure add approximately 2 5 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web hatice kocataş bursa uludağ university faculty of arts and sciences department of chemistry 16059 bursa turkey abstract in the present work a surfactant assisted convenient extraction method was developed for the isolation of trimyrustin from nutmeg

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and to a lesser extent animals especially marine animals

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web aug 4 2008 the purpose of this experiment was extract isolate and purify the natural product trimyristin from the spice nutmeg the trimyristin was extracted by using the solvent diethyl ether and

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

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web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

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web may 1 2011 experiment 3 1 february 2010 grade a small scale extraction of trimyristin from nutmeg introduction the purpose of this experiment is to collect trimyristin from an amount of nutmeg provided trimyristin is a type of triglyceride containing the fatty acid myristic acid also known as tetradecanoic acid

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