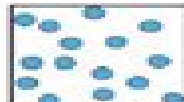


The Three States of Matter

Grade 5 - Properties and Changes in Matter

Handout - Answers

1. Draw the particles of the three different states of matter (solid, liquid, and gas).

Solid	Liquid	Gas
		

2. Provide an example of each state of matter discussed:

Solid: **Any solid (e.g. Chair)** Liquid: **Any liquid (e.g. juice)** Gas: **Any gas (e.g. Oxygen)**

3. At what temperature does water turn into a solid? **0°C**
4. At what temperature does water turn into a gas? **100°C**
5. Based on the experiments conducted do you think all liquids turn to solid at the same temperature? Do you think all liquids turn to gas at the same temperature? Explain.
(Based on any experiment) **Different liquids turn to gas at different temperatures. An example would be that the sugar crystals turned back into a solid even though it was not 0°C or below.**
6. Once something is a solid, does it stay that way forever? Explain.
***Hint: Think about the ice cream experiment and use it as an example.**
No, it can change back to being a liquid (and possibly a gas). For example, we made ice cream (a solid) out of liquid cream but when I left it out too long some of it started to melt and turn back into a liquid
7. Challenge Question: If sugar is a solid then why does it take the shape of the container it is put in? What about with salt or sand?
Sugar (as well as salt and sand) is still a solid. Each crystal has a fixed shape and volume, does not flow under shear stress, does not take the shape of its container, and is incompressible. Many small solids (grains) moving together can sometimes act like a liquid (called granular flow).

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Table of Contents Nature Of Liquids Questions Answers

1. Understanding the eBook Nature Of Liquids Questions Answers
 - The Rise of Digital Reading Nature Of Liquids Questions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Nature Of Liquids Questions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nature Of Liquids Questions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nature Of Liquids Questions Answers
 - Personalized Recommendations
 - Nature Of Liquids Questions Answers User Reviews and Ratings
 - Nature Of Liquids Questions Answers and Bestseller Lists

5. Accessing Nature Of Liquids Questions Answers Free and Paid eBooks
 - Nature Of Liquids Questions Answers Public Domain eBooks
 - Nature Of Liquids Questions Answers eBook Subscription Services
 - Nature Of Liquids Questions Answers Budget-Friendly Options
6. Navigating Nature Of Liquids Questions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Nature Of Liquids Questions Answers Compatibility with Devices
 - Nature Of Liquids Questions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nature Of Liquids Questions Answers
 - Highlighting and Note-Taking Nature Of Liquids Questions Answers
 - Interactive Elements Nature Of Liquids Questions Answers
8. Staying Engaged with Nature Of Liquids Questions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nature Of Liquids Questions Answers
9. Balancing eBooks and Physical Books Nature Of Liquids Questions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nature Of Liquids Questions Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Nature Of Liquids Questions Answers
 - Setting Reading Goals Nature Of Liquids Questions Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Nature Of Liquids Questions Answers
 - Fact-Checking eBook Content of Nature Of Liquids Questions Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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