

Purpose:
To see if
lemons make
electricity.

LEMON BATTERY

By: [REDACTED]

Results:

Test Item	Volts
1 lemon	.59
2 lemons	1.2
4 lemons	1.45
AAA battery	1.52
Potato	.54

Hypothesis:
Will more
lemons make
more volts?

Materials Used:
Lemons
Pennies
Nails
Electrical Wire
Volt meter
Potato
Triple AAA battery

Procedure:

1. squeeze the lemons
2. insert nail in one side
3. cut slit and insert penny in other side
4. attach wire from penny of one lemon to the nail of the other lemon
5. test the circuit

Conclusion:
Acid in the
lemons does
produce
electricity.

Report:

After 10 days of lemon-battery research, there is a distinct correlation between the number of lemons used and the voltage produced. After 10 days of research, the results are as follows: 1 lemon = .59 volts, 2 lemons = 1.2 volts, 4 lemons = 1.45 volts. These results are very close to the results of the AAA battery, which is 1.52 volts. The results of the potato battery are also very close to the results of the lemon battery, which is .54 volts.

Variables:

Control Group:
wire, penny, nail

Experimental Group:
lemons



Lemon Electricity Science Project Display Board

Jin-Ying Zhang



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