1.1 carbon or organic 1.2 When carbon atoms bond covalently to other carbon atoms in short or long chains or rings. 1.3 man-made 1.4 millions 1.5 Hydrocarbons are molecules consisting of carbon and hydrogen

20.1

aborns only.

T. St. Indianances.

2.2

2.3 C, H,

3.3

C H.

CH, CH, CH, CH, CH, CH,

5.1 C. Harry 5.2 Compounds in which there are no multiple bonds between C atoms in their hydrocarbon chain.

(It contains only single bonds between carbon atoms and the carbon atoms are surrounded by the maximum number of hydrogen atoms.)

5.3 unreactive 5.4 well 5.5 Inthe S. S. Acres

6.1 A series of organic compounds that can be described by the same general formula

in which one member differs from the next. with a CH, group.

6.2 yes 0.3 yes 6.4 C. H. ... 65.55

6.6 Physical properties change slightly and regularly as the chain length gets become and

6.7 Boiling points increase.

7.1 straight chain or branched chain 7.2.1

$$\frac{1}{1}$$
 $c = \frac{1}{1} - \frac{1}{1} = \frac{1}{1} - \frac{1}{1} - \frac{1}{1}$

The double bond must alternate with a single bond. The next two examples are mot conjugated double bonds.

$$H - \frac{1}{1} - \frac{1}{1} = C = \frac{1}{1} - \frac{1}{1} - H$$

7.3 delocalised

7.4 Delocalised electrons are free to move around the whole molecule, making the molecule more stable as the delocalised electrons can move further away from each other, as they can occupy the whole molecule, and thus repel each other less. Delocalised electrons do not belong to one atom, they belong to all the atoms in the empleouse.

8.1 Hydrocarbons with double bonds. 8.2.1

822

8.3 Organic molecules with the same molecular formulae, but different structural formulae.

8.4 Yes. They have different boiling exceledas.

C. H.

8.6 Compounds with one or more multiple bonds between C atoms in their hydrocarbon citations.

(The carbon atoms have multiple bonds between them and thus they are not surrounded by the maximum number of hydrogen atoms.)

19.11 Alkynes are hydrocarbons with a triple Diemid.

9.2

9.3 ethyne / acetylene 9.4 unsusturated

9.0% 10.5 10.2 the hydroxyl group -OH 100.00

10.4 CH, CH, OH

chemical behaviour.

10.5 A functional group is a bond or an atom or a group of atoms that determine(s) the physical and chemical properties of a group of organic compounds. 10.6. They all have the same functional group which is what determines their

10.7 $C_uH_uO_u \rightarrow 2C_uH_uOH + 2CO_u$ Fermentation of sugar

10.8 00,

1100.00 It is a greenhouse gas that increases global warming and climate change.

10.10 Fermenting crops to make ethanol produces CO...

The fertilizers used to grow the biofuel also produce greenhouse gases. Edible crops that are used for biofuel mean that less food will be available for people to eat.

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Yicheng Fang

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