

A man with a beard and dark hair, wearing a dark shirt, is holding a baby in his arms. The baby is sleeping peacefully. In the foreground, a lit candle with a bright flame is visible. The background is dark and slightly out of focus, showing a faint image of a woman's face.

CHRISTINA OW

Love
Reaffirmed

CANDLE LIGHT

For richer for poorer, in sickness and
in health, in death and in life

Love Reaffirmed Candle Light Book 1

Joacim Rocklöv



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 $\text{HCl aq} + \text{NaOH aq} \rightarrow \text{NaCl aq} + \text{H}_2\text{O l}$
 1 mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH aq} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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