

MATHEMATICS (SCIENCE)

Paper : II (Essay Type)

Time Allowed : 2.10 hours

Maximum Marks : 60

024-1st Annual-(10th Class)

(دوسرا گروپ)

CHR-2-24

(PART-I اول حصہ)

ریاضی (سائنس)

پرچہ : II (انشائیہ طرز)

وقت : 2.10 گھنٹے

کل نمبر : 60

12 2. Write short answers to any SIX (6) questions :

(i) Define radical equation.

(ii) Solve by factorization :

(iii) Write in standard form :

(iv) Find the discriminant of :

(v) Evaluate :

(vi) Write the quadratic equation having following roots :

(vii) Define direct variation.

(viii) If $w \propto \frac{1}{z}$ and $w = 5$ when $z = 7$ find k

(ix) Find a mean proportional between 20, 45

12 3. Write short answers to any SIX (6) questions :

(i) Define improper fraction.

(ii) If $\frac{x}{(x+a)(x-a)} = \frac{A}{x+a} + \frac{B}{x-a}$, find A and B

(iii) If $A = N$ and $B = W$, find $A - B$

(iv) If $X = \{a, b, c\}$ and $Y = \{d, e\}$, find $X \cap Y$ and $X \cup Y$

(v) If $L = \{a, b, c\}$ and $M = \{3, 4\}$, then find a binary relation of $M \times L$

(vi) Define subset and give an example.

(vii) Define range.

(viii) Write two properties of arithmetic mean.

(ix) Find harmonic mean of data :

12, 5, 8, 4

12 4. Write short answers to any SIX (6) questions :

(i) Define an angle.

(ii) Convert $\frac{\pi}{4}$ radian into degree.

(iii) Find $^{\circ}\theta^{\circ}$, when $\ell = 2.5m$, $r = 4.5m$.

(iv) Simplify each expression to a single trigonometric function $\sin^2 x \cdot \cot^2 x$

(ورق الٹئے)

2. کوئی سے چھ (6) سوالات کے مختصر جوابات لکھئے :

(i) جذری مساوات کی تعریف کیجئے۔

(ii) بذریعہ تجزیہ حل کیجئے :

(iii) مساوات کو معیاری شکل میں لکھئے :

(iv) فرق کنندہ معلوم کیجئے :

(v) قیمت معلوم کیجئے :

(vi) درج ذیل دوئس والی دو درجی مساوات لکھئے :

(vii) عکس راست کی تعریف کیجئے۔

(viii) اگر $w \propto \frac{1}{z}$ اور $w = 5$ جب $z = 7$ ہو تو k کی قیمت معلوم کیجئے۔

(ix) وسطی تناسب معلوم کیجئے :

3. کوئی سے چھ (6) سوالات کے مختصر جوابات لکھئے :

(i) غیر واجب کسر کی تعریف کیجئے۔

(ii) اگر $\frac{x}{(x+a)(x-a)} = \frac{A}{x+a} + \frac{B}{x-a}$ کی حل تلاش کیجئے۔

(iii) اگر $A = N$ اور $B = W$ اور $A - B$ معلوم کیجئے۔

(iv) اگر $X = \{a, b, c\}$ اور $Y = \{d, e\}$ اور $X \cap Y$ اور $X \cup Y$ معلوم کیجئے۔

(v) اگر $L = \{a, b, c\}$ اور $M = \{3, 4\}$ اور $M \times L$ کا ایک ثنائی ربط معلوم کیجئے۔

(vi) حقیقی سوئٹ کی تعریف کیجئے اور ایک مثال دیجئے۔

(vii) سمت کی تعریف کیجئے۔

(viii) حسابی اوسط کی دو خصوصیات لکھئے۔

(ix) مواد کا نام آجک اوسط معلوم کیجئے :

4. کوئی سے چھ (6) سوالات کے مختصر جوابات لکھئے :

(i) زاویہ کی تعریف کیجئے۔

(ii) $\frac{\pi}{4}$ rad کو ڈگری میں تبدیل کیجئے۔

(iii) $^{\circ}\theta^{\circ}$ معلوم کیجئے جبکہ $r = 4.5m$ اور $\ell = 2.5m$ ۔

(iv) جملہ کو مختصر کر کے ایک ٹرگونیٹری تعادل میں لکھئے :

$\sin^2 x \cdot \cot^2 x$

Mathematics For Class 10 Punjab Board Lahore

KJ Lindholm-Leary



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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+ \rightarrow 2\text{AgCl}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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1 $\text{K}_3\text{PO}_4 + \text{Al}(\text{NO}_3)_3 \rightarrow \text{AlPO}_4 + 3\text{KNO}_3$ ionic $3\text{PO}_4^{3-} + \text{Al}^{3+} \rightarrow \text{AlPO}_4$ s $3\text{K}^+ + 3\text{NO}_3^-$ net ionic AlPO_4 s 2 2 fec13

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq

2f aq because two nh 4 aq

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