Formula Stoichiometry Practice. This sheet should be printed, completed by hand, and scanned to produce a single pdf file with filename "lastname FormulaPract" for upload to the dropbox.

Name:

 Write down the indicated mole ratios for potassium chlorate, KClO₃. The numerator is listed first in each case.

i mol K

I mal K

moles K* to moles CIO₅.

| mol K
| mol CEO₅.

l mai CP 3 mai O

i mai Cl

3 mol D 1 mol K

3 mol 0

1 mai CLOs

Calculate molar masses for the two compounds below making sure to show your work.

M= 39.098 (0= 35.45 0= 15.999 13 = 47.997 aluminum sulfate, Ala(SOL);
Al - 26, 982 * 2 = 53,964
504 - [[12998:4] + 32.06] 3 - 286.166
342.132 9/mol

Mixed Stoich Practice

Jack Hall Westbrook, Robert Louis Fleischer

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