

CLASS SET

Mixed Stoichiometry Problems

1. How many moles of H_2 would be required to completely react with O_2 to produce 5 moles of water? **5 mol H_2**
2. $\text{H}_2\text{SO}_4 + \text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$
 - a. Balance this equation
 - b. What mass of H_2SO_4 would be required to react with 0.75 mol of NaOH ? **37g**
3. What mass of NO_2 is formed when NO reacts with 384 g of O_2 ? **1104g**
4. If 30 g of H_2SO_4 reacts with aluminum hydroxide in a double replacement reaction, what mass of water is produced? **11.0g**
5. If 10 L of CH_4 combusts with O_2 , what volume of water will be produced? **20 L H_2O**
6. Tin(II) fluoride, SnF_2 , is used in some toothpastes. It is made by the reaction of tin with hydrogen fluoride according to the following equation.
$$\text{Sn} + 2\text{HF} \rightarrow \text{SnF}_2 + \text{H}_2$$

How many grams of SnF_2 are produced from the reaction of 30 g of HF with Sn ? **118 g**
7. In a spacecraft, the carbon dioxide exhaled by astronauts can be removed by its reaction with lithium hydroxide, LiOH , according to the following chemical equation.
$$\text{CO}_2(\text{g}) + 2\text{LiOH}(\text{s}) \rightarrow \text{Li}_2\text{CO}_3(\text{s}) + \text{H}_2\text{O}(\text{l})$$

If 448 L of CO_2 is exhaled, the average amount exhaled by a person each day, how much (in grams) $\text{Li}_2\text{CO}_3(\text{s})$ is produced? **1480g Li_2CO_3**
8. When 9.8g of aluminum oxide decomposes, how many grams of aluminum metal are produced? **5.2 g Al**
9. How many grams of iodine are produced when 0.72 mol of fluorine react with potassium iodide?
10. How many grams of sodium are required to react with water to produce 5.0g of sodium hydroxide? ($\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$)
11. How many grams of H_2 are produced in the reaction in #10?
12. What volume would be occupied by the resulting hydrogen at STP?
13. 20.0g of K react with H_2O to produce KOH and H_2 gas. How many liters would the hydrogen gas occupy at STP?
14. If 30.2g of Al react with HCl to produce AlCl_3 and H_2 gas, how many liters of H_2 are produced at STP?
15. How many liters of HCl are produced by the reaction of 5.7L of H_2 with Cl_2 ?

Mixed Stoichiometry Word Problems And Answers

John C. Kotz



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Scientific and Technical Aerospace Reports* , 1985

Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 Part of the Prentice Hall Series in Educational Innovation for Chemistry this unique book is a collection of information examples and references on learning theory teaching methods and pedagogical issues related to teaching chemistry to college students In the last several years there has been considerable activity and research in chemical education and the materials in this book integrate the latest developments in chemistry Each chapter is written by a chemist

who has some expertise in the specific technique discussed has done some research on the technique and has applied the technique in a chemistry course *Chemistry Homework* Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more *Theoretical and Technological Aspects of Crystal Growth* R. Fornari, C. Paorici, 1998-03-30 Proceedings of the 10th International Summer School on Crystal Growth Rimini Italy June 1998 *Reactor Technology*, 1972 **Methods in Ecosystem Science** Osvaldo E. Sala, Robert B. Jackson, Harold A. Mooney, Robert Howarth, 2000-08-11 Ecology at the ecosystem level has both necessitated and benefited from new methods and technologies as well as those adapted from other disciplines With the ascendancy of ecosystem science and management the need has arisen for a comprehensive treatment of techniques used in this rapidly growing field *Methods in Ecosystem Science* answers that need by synthesizing the advantages disadvantages and tradeoffs associated with the most commonly used techniques in both aquatic and terrestrial research The book is divided into sections addressing carbon and energy dynamics nutrient and water dynamics manipulative ecosystem experiments and tools to synthesize our understanding of ecosystems Detailed information about various methods will help researchers choose the most appropriate methods for their particular studies Prominent scientists discuss how tools from a variety of disciplines can be used in ecosystem science at different scales *The Chemical World* John C. Kotz, 1994 *Stanford Bulletin*, 2003

The Software Finder, 1983 **English Mechanic and Mirror of Science**, 1892 **American Druggists' Circular and Chemical Gazette**, 1899 **Teaching Science for Understanding** James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities *Designing the Perfect Resume* Pat Criscito, 2005-11-01 Job seekers will find step by step instructions for creating an effective r sum that presents their background and qualifications in the very best light They will learn the difference between functional and reverse chronological r sum s and get advice on which style is best for their needs They ll also find tips on making the most effective use of type fonts graphic devices and attractive layout when preparing a r sum as well as advice on creating electronic r sum s Many different job categories are reflected in this book s 220 sample r sum s which can be adapted and personalized for use by job seekers New in this are more than 100 new r sum designs created by members of the National R sum Writers Association NRWA **The Publishers Weekly**, 1960 **Science** John Michels (Journalist), 2010

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