

## Unit 6 – Worksheet 3

### Ionic Compounds

#### Properties

#### Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- a.  $\text{Na}_2\text{O}$
- b.  $\text{K}_2\text{S}$
- c.  $\text{MgCl}_2$
- d.  $\text{CaBr}_2$
- e.  $\text{BaI}_2$
- f.  $\text{Al}_2\text{S}_3$
- g.  $\text{CsBr}$
- h.  $\text{AgF}$

2. Give the name of the following simple binary ionic compounds.

- a.  $\text{Na}_3\text{N}$
- b.  $\text{K}_2\text{O}$
- c.  $\text{AgBr}$
- d.  $\text{MgI}_2$
- e.  $\text{SrO}$

3. Write the formula for the following binary ionic compounds.

- a. lithium bromide
- b. sodium iodide
- c. silver sulfide
- d. cesium oxide
- d. beryllium iodide
- f. barium hydride
- g. aluminum fluoride
- h. potassium oxide

4. Write the formula for these ionic substances.

- a. silver oxide
- b. aluminum sulfide
- c. sodium nitride
- d. barium chloride
- e. strontium hydride

# Modeling Chemistry U6 Ws 4 V2 Answers

**RJ Alexander**



## **Modeling Chemistry U6 Ws 4 V2 Answers:**

## **Modeling Chemistry U6 Ws 4 V2 Answers** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Modeling Chemistry U6 Ws 4 V2 Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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