

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

1. Calculate the number of moles of potassium chlorate, KClO_3 (s), that must decompose to produce potassium chloride, KCl (s), and 1.8 moles of oxygen gas.

Equation:	$2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$			1.2 moles KClO_3 needed
Before:	1.2	0	0	
Change:	-1.2	+1.2	+1.8	
After:	0	1.2	1.8	

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equation:	$\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$				0.200 moles HCl needed
Before:	0.100	?	0	0	
Change:	-0.100	-0.200 +	0.100 +	0.100	
After:	0	?	0.100	0.100	

3. Ethane, C_2H_6 reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	$2 \text{C}_2\text{H}_6 + 7 \text{O}_2 \rightarrow 4 \text{CO}_2 + 6 \text{H}_2\text{O}$			
Before:	2.20	xs	0	0
Change:	-2.20	-7.70	+4.40	+6.60
After:	0	xs	4.40	6.60

$$\text{O}_2 \text{ needed: } 7.70 \text{ mole} \times \frac{32.0 \text{ g}}{1 \text{ mole}} = 246 \text{ g O}_2 \text{ required}$$

4. Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation:



$$0.73 \text{ g} \times \frac{1 \text{ mole}}{183 \text{ g}} = .0040 \text{ mole}$$

Equation:	$\text{Ni}(\text{NO}_3)_2 + 2 \text{NaOH} \rightarrow \text{Ni}(\text{OH})_2 + 2 \text{NaNO}_3$			
Before:	0.0040	XS	0	0
Change:	-0.0040	-0.0080	+0.0040	+0.0080
After:	0	XS	0.0040	0.0080

$$\text{NaNO}_3 \text{ produced: } 0.0080 \text{ mole} \times \frac{85.0 \text{ g}}{1 \text{ mole}} = 0.68 \text{ g}$$

Modeling Chemistry Unit 3 Answers

RJ Shavelson



Modeling Chemistry Unit 3 Answers:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11
Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Towards Sustainable and Scalable Educational Innovations Informed by the Learning Sciences** Chee-Kit Looi, David H. Jonassen, Mitsuru Ikeda, 2005 One of the basic principles that underpin the learning sciences is to improve theories of learning through the design of powerful learning environments that can foster meaningful learning Learning sciences researchers prefer to research learning in authentic contexts This book focuses on learning sciences in the Asia Pacific context Resources in Education, 1997-04 **Chemistry in the Community (Enhanced Core Four)** American Chemical Society, 2006-02-15 Strategies Harold David Schmidt, 1988

Macmillan Encyclopedia of Chemistry J. J. Lagowski, 1997 Comprehensive and up to date this unique four volume set offers readers a complete overview of the broad spectrum of general chemistry It enables them to obtain a basic yet thorough understanding of matter the processes it undergoes the principles that govern it and the international cast of men and women who have been critical in the development of the science of chemistry From elements atoms and molecules to terochemistry spectroscopy and chemical bonding its clear and concise explanations provide an illuminating and readily comprehensible introduction Key presentations include forty element definition articles each providing basic periodic table information and general information on the element in question Ninety five biographical articles deal with prominent chemists while other articles provide additional historical context particularly with respect to eighteenth nineteenth and twentieth century developments *Learning Progressions in Science* Alicia C. Alonzo, Amelia Wenk Gotwals, 2012-07-30
Learning progressions descriptions of increasingly sophisticated ways of thinking about or understanding a topic National Research Council 2007 represent a promising framework for developing organized curricula and meaningful assessments in science In addition well grounded learning progressions may allow for coherence between cognitive models of how understanding develops in a given domain classroom instruction professional development and classroom and large scale assessments Because of the promise that learning progressions hold for bringing organization and structure to often disconnected views of how to teach and assess science they are rapidly gaining popularity in the science education community However there are signi cant challenges faced by all engaged in this work In June 2009 science education researchers and practitioners as well as scientists psychometricians and assessment specialists convened to discuss these challenges as part of the Learning Progressions in Science LeaPS conference The LeaPS conference provided a structured forum for considering design decisions entailed in four aspects of work on learning progressions de ning learning

progressions developing assessments to elicit student responses relative to learning progressions modeling and interpreting student performance with respect to a learning progressions and using learning progressions to influence standards curricula and teacher education This book presents specific examples of learning progression work and syntheses of ideas from these examples and discussions at the LeaPS conference *Resources in Education*, 1986 **Earth Science**, 2001 *Title List of Documents Made Publicly Available*, 1987 *The Educational Review*, 1912 **Holt Chemistry** Ralph Thomas Myers, 2004 *Scientific American*, 1874 **Fibre & Fabric**, 1927 *Normal Instructor and Primary Plans*, 1914 *The United States Catalog* Mary Burnham, Carol Hurd, 1928 **Biology** Neil A. Campbell, Jane B. Reece, 2005 Neil Campbell and Jane Reece's BIOLOGY remains unsurpassed as the most successful majors biology textbook in the world This text has invited more than 4 million students into the study of this dynamic and essential discipline The authors have restructured each chapter around a conceptual framework of five or six big ideas An Overview draws students in and sets the stage for the rest of the chapter each numbered Concept Head announces the beginning of a new concept and Concept Check questions at the end of each chapter encourage students to assess their mastery of a given concept New Inquiry Figures focus students on the experimental process and new Research Method Figures illustrate important techniques in biology Each chapter ends with a Scientific Inquiry Question that asks students to apply scientific investigation skills to the content of the chapter

Encyclopedia of Environmental Analysis and Remediation, Volume 1 Robert A. Meyers, 1998-03-23 This encyclopedia is composed of an eight volume set that provides an overview of the field of environmental analysis The contents are divided into major content areas including air pollution control environmental law and environmental sampling The volumes are organized alphabetically with each article signed by the author's The individual articles begin with a summary of the topic heading and then divide the text into subtopics indicated by boldface headings The articles are written clearly however the authors assume a basic knowledge of chemistry and math on the part of the reader For example the acid mine drainage article refers to the Arrhenius equation but does not clarify this statement in the text Each article contains graphs as well as pictures to illustrate points made in the text The articles are long and provide a detailed explanation of each topic The authors also provide a bibliography at the end of each article Special features of the encyclopedia include a list of contributors a table of conversion factors and a list of abbreviations and acronyms The preface outlines the general contents of the encyclopedia The preface also includes sections that suggest the target audience and recommended usages for the set The final volume of the set contains an alphabetic index to the topics contained in the volumes *CEP Software Directory*, 1995 *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office, 1963

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