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Chemistry – Unit 10 Worksheet 3

1. Element X has two natural isotopes. The isotope with a mass number of 6 has a relative abundance of 7.5%. The isotope with a mass number of 7 has a relative abundance of 92.5%. Determine the average molar mass for the element from these figures. What is the true identity and atomic number of element X?
2. The element copper is found to contain the naturally occurring isotopes $^{63}_{29}\text{Cu}$ and $^{65}_{29}\text{Cu}$. The relative abundances are 69.1% and 30.9% respectively. Calculate the average molar mass of copper.
3. Uranium has three isotopes with the following percent abundances: $^{234}_{92}\text{U}$ (0.0058%), $^{235}_{92}\text{U}$ (0.71%), $^{238}_{92}\text{U}$ (99.23%). What do you expect the molar mass of uranium to be in whole numbers? Why?
4. A sample of silver as it occurs in nature is 52.0% of isotope $^{107}_{47}\text{Ag}$ and 48.0% of isotope $^{109}_{47}\text{Ag}$. What is the average molar mass of silver? (Compare your result with the value given in the periodic table).

Modeling Chemistry Unit 10 2 Answer

**Václav Snášel, Ajith Abraham, Emilio S.
Corchado**



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Soft Computing Models in Industrial and Environmental Applications Václav Snášel, Ajith Abraham, Emilio S. Corchado, 2012-08-23 This volume of Advances in Intelligent and Soft Computing contains accepted papers presented at SOCO 2012 held in the beautiful and historic city of Ostrava Czech Republic in September 2012 Soft computing represents a collection or set of computational techniques in machine learning computer science and some engineering disciplines which investigate simulate and analyze very complex issues and phenomena After a thorough peer review process the SOCO 2012 International Program Committee selected 75 papers which are published in these conference proceedings and represents an acceptance rate of 38% In this relevant edition a special emphasis was put on the organization of special sessions Three

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Handbook of Soil Sciences Pan Ming Huang, Yuncong Li, Malcolm E. Sumner, 2011-11-17 An evolving living organic inorganic covering soil is in dynamic equilibrium with the atmosphere above the biosphere within and the geology below It acts as an anchor for roots a purveyor of water and nutrients a residence for a vast community of microorganisms and animals a sanitizer of the environment and a source of raw materials for co

Concepts and Models of Inorganic Chemistry, Problems Bodie E. Douglas, Darl H. McDaniel, John J. Alexander, 1983-08-31 A clear introduction to modern inorganic chemistry Covering both theory and descriptive chemistry the text begins with atomic structure bonding and stereochemistry and then treats inorganic solids acids and bases and bioinorganic chemistry This second edition includes optional sections on group theory very thorough discussions of inorganic solids and expanded material on subjects such as the mechanisms of reactions and bioinorganic chemistry Presents numerous figures to encourage model thinking and provides solved examples

Models in Chemical Science George Simms Hammond, 1971

Modelling in Aquatic Chemistry OECD Nuclear Energy Agency, 1997

Boston Medical and Surgical Journal, 1929

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The Chemical News, 1882

British Chemical Abstracts, 1928

Journal of the South African Institute of Mining and Metallurgy South African Institute of Mining and Metallurgy, 1912

The Industrial Arts Index, 1927

ERDA Energy Research Abstracts, 1985

Australian Journal of Chemistry, 2006

Russian Journal

of Inorganic Chemistry ,1995

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