

## Chemistry – Unit 5 Worksheet 2

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

$$3.3 \text{ g} \times \frac{1 \text{ mole}}{63.5 \text{ g}} = 5.2 \times 10^{-2} \text{ mol or } 0.052 \text{ mol}$$

2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

$$4.42 \text{ g} \times \frac{1 \text{ mole}}{55.8 \text{ g}} = 7.92 \times 10^{-2} \text{ mol or } 0.0792 \text{ mol}$$

3. A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

$$0.050 \text{ g} \times \frac{1 \text{ mole}}{18.0 \text{ g}} = 2.78 \times 10^{-3} \text{ mol} \rightarrow 2.8 \times 10^{-3} \text{ mol}$$

4. What mass of water would you need to have 15.0 moles of H<sub>2</sub>O?

$$15.0 \text{ moles} \times \frac{18.0 \text{ g}}{1 \text{ mole}} = 2.70 \times 10^2 \text{ g}$$

5. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

$$737 \text{ g} \times \frac{1 \text{ mole}}{58.5 \text{ g}} = 12.6 \text{ mol}$$

|    |              |
|----|--------------|
| Na | 23.0         |
| Cl | 35.5         |
|    | <hr/> 58.5 g |

6. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

$$0.050 \text{ moles} \times \frac{84.0 \text{ g}}{1 \text{ mole}} = 4.2 \text{ g}$$

|    |            |
|----|------------|
| Na | 23.0       |
| H  | 1.0        |
| C  | 12.0       |
| O  | 48.0       |
|    | <hr/> 84.0 |

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2.50 kg of rust?

$$2.50 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mole}}{159.6 \text{ g}} = 15.7 \text{ mol Fe}_2\text{O}_3$$

|    |             |
|----|-------------|
| Fe | 111.6       |
| O  | 48.0        |
|    | <hr/> 159.6 |

# Modeling Chemistry Unit 2 Quiz Answers

**Michael Brown**



## Modeling Chemistry Unit 2 Quiz Answers:

**Resources in Education** ,1980      Title List of Documents Made Publicly Available U.S. Nuclear Regulatory Commission,1987      *Prentice Hall Chemistry* ,2000      **NASA Technical Memorandum** ,1963      **Spacecraft Propulsion** ,1993      **Pharmaceutics [GPAT] - Books [Study Notes] 7 in 1 Books with 2500+ Question Answer As Per Updated Syllabus** DIWAKAR EDUCATION HUB,2022-04-01 Pharmaceutics GPAT Books Study Notes 7 Books with 2500 Question Answer As Per Updated Syllabus Design by Expert Faculties for Secure 152 Marks in Graduate Pharmacy Aptitude Test Asked 38 MCQ in Exam Highlights of Books As Per Updated Syllabus Graduate Pharmacy Aptitude Test 7 Booklets theory MCQ In Each Book given 4 Chapters in Details Total 28 Covered all 28 Chapters Ex Pharmacy Profession Introduction to Pharmaceuticals Introduction to dosage form Sources of drug information Total 2500 Questions Answer Numerical with Explanation Design by Pharma Professor Topper Qualified Students Total 7 Booklets For Secured 152 Marks in Exam For More Details Call Whats App 7310762592 7078549303      *Energy Research Abstracts* ,1986      **Transactions of the American Nuclear Society** ,1984      *EPA Publications Bibliography Quarterly Abstract Bulletin* United States. Environmental Protection Agency,1993-07      **38th AIAA/ASME/SAE/ASEE Joint Propulsion Conference & Exhibit: 02-3850 - 02-3899** ,2002      *Resources in Education* ,1982      *INIS Atomindex* ,1995      **Government Reports Announcements & Index** ,1996      *Abstracts* ,1989      **The British National Bibliography** Arthur James Wells,2002      *Government Reports Annual Index* ,1992      The Impact of Formative Assessment Techniques on the Instruction of the High School Biology Units of Photosynthesis and Cellular Respiration Shanna Fawn Tury,2010      **Science Abstracts** ,1992

**The Proceedings of the ... International Conference on Fluidized-Bed Combustion** ,1987      Evaluation of Fly Ash Quality Control Tools Tyson David Rupnow,2010 Many entities currently use fly ash in portland cement concrete PCC pavements and structures Although the body of knowledge is great concerning the use of fly ash several projects per year are subject to poor performance where fly ash is named as the culprit The main focus of the research was to identify penetration type devices and test procedures including the Iowa Set Time Test Gillmore needle and Vicat needle Another focus of the investigation was the quick heat generation index test For the first objective three penetration type test devices were investigated including the Vicat needle Gillmore needle and the pocket penetrometer Class C fly ash samples were obtained from about ten sources available to Louisiana Department of Transportation and Development LADOTD The second objective was to identify if the quick heat generation test can identify small changes in class C fly ash whether that be a change in chemistry or a physical change in the fly ash fineness Statistical modeling was used to determine if a relationship existed between the various initial and final set times and the maximum temperature of the fly ash paste and the fly ash chemistry and fineness

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