

Chemistry Unit 8

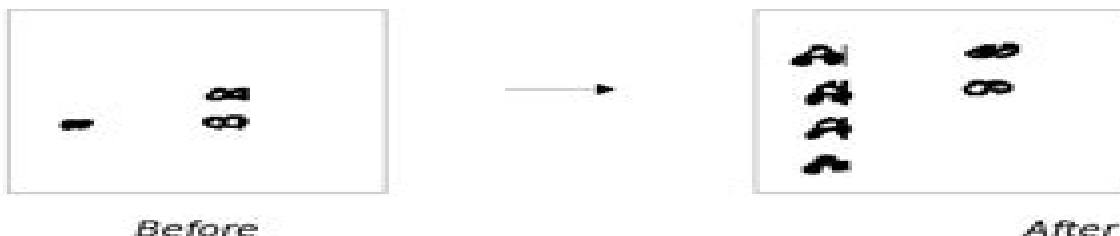
Worksheet 3: Adjusting to Reality - Limiting Reactant

1. Write the balanced equation for the reaction between hydrogen and oxygen.

Balanced Equation: $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$

Suppose that 4 molecules of hydrogen gas and 4 molecules of oxygen gas react to form water.

Make a drawing that represents the reaction container before and after the reaction.



- 4 How many molecules of water can be produced?

Oxygen, Leftover molecules Which reactant is in excess? Why?

- 2 How many molecules of excess reactant are there?

Construct a Before-Change-After Table for this reactant mixture:

Bal. Equation: $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$

Before:	4	4	0
Change:	-4	-2	+4
After:	0	2	4

According to the table you just made,

- 4 How many molecules of water can be produced?

Oxygen, 2 molecules left over post-reaction Which reactant is in excess? Why?

- 2 How many molecules of excess reactant are there?

Based on your two methods of analysis above, what determines how much product can be made from a particular reactant mix?

The determining factor is which of the reactants runs out first. This determines how much product can be made, even if another reactant is in excess.

Modeling Chemistry Unit 6 Test

Minjie Lin



Modeling Chemistry Unit 6 Test:

SME Mineral Processing and Extractive Metallurgy Handbook Courtney A. Young, 2019-02-01 This landmark publication distills the body of knowledge that characterizes mineral processing and extractive metallurgy as disciplinary fields. It will inspire and inform current and future generations of minerals and metallurgy professionals. Mineral processing and extractive metallurgy are atypical disciplines requiring a combination of knowledge, experience, and art. Investing in this trove of valuable information is a must for all those involved in the industry: students, engineers, mill managers, and operators. More than 192 internationally recognized experts have contributed to the handbook's 128 thought-provoking chapters that examine nearly every aspect of mineral processing and extractive metallurgy. This inclusive reference addresses the magnitude of traditional industry topics and also addresses the new technologies and important cultural and social issues that are important today.

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