Name: \_\_\_ Date: Mods:

#### Honors Chemistry Midterm: Review Packet

#### PART 1

Write in the number of significant figures for each of the following measurements:

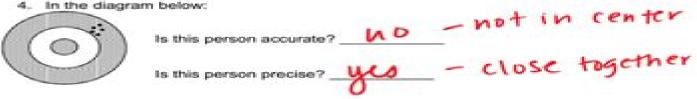
3 0.00290 g 4 50.00 m 2 0.070 kg /

- Using the following masses: 0.0562 kg, 124.213 g and 1635 mg.
  The answer should be reported as: \_\_\_\_\_\_g The answer should be reported as: \_\_\_\_\_\_g 1.635g
- 3. a) A 5.75 mL sample of mercury has a measured mass of 77.05 g. The density is 13.49/mL

b) Mercury's accepted density is 13.53 g/mL. The % error in your measurement is: . 961 70

acture ×100 113.639/ms -13.49/ms ×100

In the diagram below:



- 5. Identify the following as a chemical change (CC), physical change (PC), chemical property (CP), or physical property (PP):
  - a) burning wood CC

b) mass of wood PP

c) flammability of wood CP d) chopping up wood PC

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## **Table of Contents Mcps Honors Chemistry B Review Packet Answers**

- 1. Understanding the eBook Mcps Honors Chemistry B Review Packet Answers
  - The Rise of Digital Reading Mcps Honors Chemistry B Review Packet Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Mcps Honors Chemistry B Review Packet Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mcps Honors Chemistry B Review Packet Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mcps Honors Chemistry B Review Packet Answers
  - Personalized Recommendations
  - Mcps Honors Chemistry B Review Packet Answers User Reviews and Ratings
  - Mcps Honors Chemistry B Review Packet Answers and Bestseller Lists

- 5. Accessing Mcps Honors Chemistry B Review Packet Answers Free and Paid eBooks
  - Mcps Honors Chemistry B Review Packet Answers Public Domain eBooks
  - Mcps Honors Chemistry B Review Packet Answers eBook Subscription Services
  - Mcps Honors Chemistry B Review Packet Answers Budget-Friendly Options
- 6. Navigating Mcps Honors Chemistry B Review Packet Answers eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Mcps Honors Chemistry B Review Packet Answers Compatibility with Devices
  - Mcps Honors Chemistry B Review Packet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mcps Honors Chemistry B Review Packet Answers
  - Highlighting and Note-Taking Mcps Honors Chemistry B Review Packet Answers
  - Interactive Elements Mcps Honors Chemistry B Review Packet Answers
- 8. Staying Engaged with Mcps Honors Chemistry B Review Packet Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mcps Honors Chemistry B Review Packet Answers
- 9. Balancing eBooks and Physical Books Mcps Honors Chemistry B Review Packet Answers
  - ∘ Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mcps Honors Chemistry B Review Packet Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Mcps Honors Chemistry B Review Packet Answers
  - Setting Reading Goals Mcps Honors Chemistry B Review Packet Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcps Honors Chemistry B Review Packet Answers
  - Fact-Checking eBook Content of Mcps Honors Chemistry B Review Packet Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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