

Chemistry – Unit 5 Worksheet 1

Relative Masses of the Elements

Relative Mass From Counts

We have established that atoms have different relative masses. We can use the relative mass of each element to find the relative mass of a compound.

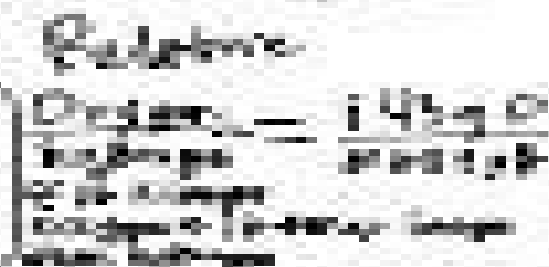
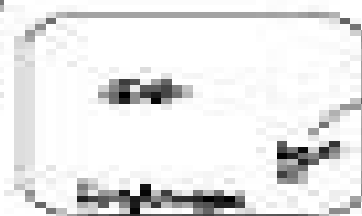
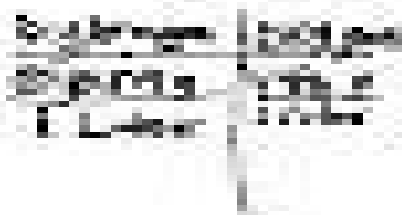
1. Equal numbers of atoms contain the same number of nucleons as the same number of protons.

Example: **Hydrogen Molecule**

2. Atoms of the same element have the same relative mass.

H₂, O₂, N₂, Cl₂, Br₂

3. The relative mass of a molecule is the sum of the relative masses of the atoms in it. For example, the relative mass of a water molecule (H₂O) is 18. This is because it has two hydrogen atoms (each with a relative mass of 1) and one oxygen atom (with a relative mass of 16).



4. We can use the relative mass of a molecule to find the relative mass of each of the elements in it. For example, the relative mass of a water molecule is 18.02. The relative mass of the oxygen in it is 16.00. This means that the relative mass of the oxygen in a water molecule is 16.00/18.02 = 0.888. This means that the relative mass of the oxygen in a water molecule is 0.888 times the relative mass of the water molecule.

Relative Mass From Composition

Many substances are made up of two or more different elements. For example, water is made up of hydrogen and oxygen. The relative mass of a substance can be found by adding the relative masses of the elements in it. For example, the relative mass of water is 18.02. This is because it has two hydrogen atoms (each with a relative mass of 1) and one oxygen atom (with a relative mass of 16).

5. If we know the relative mass of a substance and the relative mass of each of the elements in it, we can find the relative mass of each of the elements in it. For example, the relative mass of water is 18.02. The relative mass of the oxygen in it is 16.00. This means that the relative mass of the oxygen in a water molecule is 16.00/18.02 = 0.888. This means that the relative mass of the oxygen in a water molecule is 0.888 times the relative mass of the water molecule.



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