

General Chemistry - Unit 1 (Part 1) Review WS - **ANSWERS**

Directions: Complete the following questions on a separate piece of paper.

LT #2: Understand the experiments and the historical theories that lead to the modern atomic theory.

- Describe Dalton's beliefs about the atom and chemical reactions with atoms.
All elements are composed of atoms; all atoms of a given element are identical, that atoms of different elements are different; chemical compounds are formed when atoms combine; atoms are not created or destroyed in chemical reactions (indivisible); solid sphere.
- Describe the atom according to Thomson's theory.
Thomson found that the atom consisted of a positive mass but there were negative pieces scattered throughout the positive area (Plum Pudding Model). He discovered the electron (negatively charged).
- Describe Rutherford's gold foil experiment. How did he change the model of the atom after the experiment?
Rutherford discovered the nucleus contained protons by shooting alpha particles (+) through gold foil. He found that most went through but some bounced back. He found the nucleus, which is very, very small and dense is found in the center of the atom, with the electrons around the outside.
- What did Chadwick change about the atomic model?
Chadwick (along with the Curies) found that the nucleus also contained the neutrons, which have approximately the same mass as the proton, but have no charge.

LT #3: Understand how subatomic particles determine the properties of an atom or ion.

- List the three subatomic particles, their location, their charge and their relative mass.
*Proton—found in nucleus, + charge, 1 amu (atomic mass unit)
Neutron—found in nucleus, no charge, 1 amu
Electron—cloud surrounding the nucleus, - charge, approx. no mass*
- What is a cation and how is it formed?
A cation is a positive ion and is formed when an atom loses one or more electrons.
 - What is an anion and how is it formed?
An anion is a negative ion and is formed when an atom gains one or more electrons.
- Complete the following charts for atoms and ions:

ATOM	ISOTOPE SYMBOL	ATOMIC #	MASS #	#PROTONS	#ELECTRONS	#NEUTRONS
Sodium - 23	²³ ₁₁ Na	11	23	11	11	12
Aluminum - 27	²⁷ ₁₃ Al	13	27	13	13	14
Chlorine - 35	³⁵ ₁₇ Cl	17	35	17	17	18
Barium - 136	¹³⁶ ₅₆ Ba	56	136	56	56	80

SYMBOL	MASS	#PROTONS	#ELECTRONS	#NEUTRONS	NET CHARGE
K ⁺	39	19	18	20	+1
Br ⁻	80	35	36	45	-1
Fe ³⁺	56	26	23	30	+3
Se ²⁻	79	34	36	45	-2

- Are all atoms of the same element identical? If not, how can they differ?
No; isotopes differ by # of neutrons. (Isotopes have the same # of protons and electrons.)
- Which of lead's four isotopes is most abundant: lead-204, lead-206, lead-207 or lead-208? Explain how you know which isotope is more abundant. *The average atomic mass for lead is 207.2 amu; therefore, the most abundant isotope is lead-207*

LT #4: Apply the mole concept (molar mass, mole equalities, conversions/dimensional analysis).

- Calculate the molar mass of each of the following atoms or compounds:
 - zinc
65.4 g/mol
 - barium phosphate, Ba₃(PO₄)₂
3(137.3) + 2(31.0) + 8(16.0) = 601.9 g/mol

Modeling Chemistry U6 Test Answers

Roman Wölfel



Modeling Chemistry U6 Test Answers:

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web tomie kawakami better known mononymously as tomie is a character from the japanese horror manga and film series of the same name created by junji ito tomie was introduced in ito s 1987 manga tomie which was published in

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web tarihi uzay vampir yaoi yuri anılar hem iyi hem de kötü olarak lise tecrübelerimi kaplıyor arkadaşlarıyla takılmak ya da sınavlardan kafanı kaldıramamak herkes o zamanlardan bir şeyler hatırlar belli bir lisenin bir sınıfı lise günlerine üzüntüyle bakmalarını sağlayacak bir o

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web 15 temmuz günceli ibitsu ya devam ediyoruz bu seriye benim suşi manga Çeviri yi kurma düşüncelerim varken başlanılmıştı tabi düşünce aşamasında kaldı buradan devam ediyoruz aynı çevirmen ve editorle nea manga Çeviri ile koroshiya 1 ichi the killer a başladık başlar başlamaz da ben ortadan kayboluyorum 1

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web tomie is a manipulative hedonistic and extremely selfish woman seducing men to do her bidding she is one of junji ito s most recurring characters appearing for a grand total of 3 series tomie tomie part 2 and tomie again

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web dec 21 2021 tomie is a story where people fascinated with the girl end up killing her accelerating her rebirth the proliferation of tomie was created while writing a serial storyline which helped greatly to convey the concept of regeneration

itou junji collection tomie myanimelist net - Jun 19 2023

web apr 27 2018 the deceased is not just any student she s tomie kawakami a popular girl with an almost otherworldly beauty her death was particularly gruesome her body was dismembered and the pieces scattered as the class tries to make sense of the situation they are shocked when a familiar voice calls out to them from the doorway apologizing

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web feb 7 2011 in a high school classroom students mourn the loss of one of their own tomie kawakami who has been murdered and dismembered shocked by the announcement of her death the class is puzzled by the cruel fate that has befallen someone so dear to them such a radiant and beautiful girl did not deserve such a hideous demise

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web the series focuses on the titular tomie kawakami a beautiful young girl identified by a mole under her left eye who drives her stricken admirers to madness often resulting in her own death however due to her ability of regeneration she comes back

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web these rather disturbing tales tell the story of a high school girl named tomie who can be seen as a living embodiment of lust and all the negative emotions that go along with it such as jealousy tomie is the ultimate self destructive entity yet ironically she survives anything

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web mar 17 2022 tomie s full name is tomie kawakumi she makes repeated appearances throughout the three volumes of junji ito s tomie she is described as a young woman with beautiful hair and having a mole underneath her eye

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