

## Phet - KINETIC MOLECULAR THEORY

1. By visiting the link given in lab description, Starting the "Intro" screen and checking the box for "Collision Counter" is checked.
2. Blue particles are Heavy while red particles are light
3. Some predictions related to the lab are
  - i. The increase in temperature is directly related to the pressure, if one keeps number of particles same in the container.
  - ii. If one keeps temperature and volume same in the container, then addition of number of particles have direct relation with the pressure.
  - iii. When heavy and light particles are exposed to the same conditions, they both behaves in the same manner for pressure, temperature and volume. However, they show a large difference in the case of wall collision due to their speed differences. Small particles move with higher speed as compare to heavy particles.
4. By putting one pump of heavy (blue) gas particles into the container at room temperature and observation for sometime gives information for the answer of questions given below.

- a. Describe how the particles behave over a span of about 30 seconds. Do the particles ever stop moving?

**Answer:**

The particles randomly move in the container strike with its walls. The parameters like temperature, pressure and volume have constant value for entire period of 30 seconds. Moreover, the particles remain in motion over the entire period of time and never come to rest.

- b. What is the Count for wall collisions?

**Answer:**

The value of wall count comes out 34 as shown in figure 1. Count for wall collisions is obtain by hitting the green play button on the "Wall Collision" box in the top left. It will run for 10 picoseconds; this value is also recorded in data table 1 under room temperature.

- c. What is pressure reading in atm?

**Answer:**

The pressure reading in gauge was 5.8 atm as shown in figure 1.



Figure 1. Wall collision and Pressure reading at room temperature

- d. Describe the motion of the particles in the data table at this point in time.

**Answer:**

Particle moves randomly everywhere with wall collision value of 34. They remain in continuous motion and never comes to rest conditions. Also, motion is described in data table 1 under room temperature.

# Kinetic Molecular Theory Answers Key

**CO Houle**



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