Principle.		

#### COVALENT BOND WEBQUEST

Section.

DRECTIONS: Answer the following by going to the notes website. You can go to Mrs. Showlati's website and use the links or type in the addresses for each section if needed.

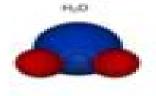
#### Activity #1- Introduction to Covalent Bonding.

Open Chemical Bonding Scroll down to the heading "Covalent Bonding." OR (http://www.visconiesrning.com/library/module\_viewer.prp?mst-55)

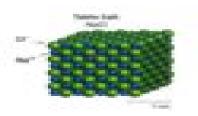
1. As opposed to		bonding in which a -	complete transfer of electrons	
occur.	bonding occurs when two or more elements			
	electrons, Gavale	and bioinding occurs beca	ause the atoms in the compound	
have a similar tendency	y for electrons (pr	enenally to	electrons). This	
most commonly occurs when two		bon	d together. Because both of the	
nonmetals will want to		electrons, the elements involved will share		
electrons in an effort to		their valence shells.		
3. How many valence s	nlectrons does hy	drogen need to have a	full first shell?	
4. How does the hydro		1050 		
5. What compound doe	ss hydrogen form	7		
6. How do hydrogen as				

8. Label the substances below as "lonic" or "covalent".

 Visit the simulation. Covatent bonding between hydrogen atoms. OR http://www.visionleaming.com/library/flash\_viewer.php?coi+13466mid+55



Describe or draw what you see.



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