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Ionic bonding practice

How many valence electrons do each of the following elements have?

- 1) carbon _____
- 2) selenium _____
- 3) xenon _____
- 4) potassium _____

Which of the following ions are likely to be formed? Circle one:

- | | | | | | |
|-------------|-----|----|---------------|-----|----|
| 5) N^{+5} | yes | no | 8) Al^{+2} | yes | no |
| 6) He^{+} | yes | no | 9) P^{-3} | yes | no |
| 7) F^{-1} | yes | no | 10) Mg^{+2} | yes | no |

Write the formula which result from combining the following ions:

- 11) K^{+1} and Br^{-} _____
- 12) Li^{+1} and O^{-2} _____
- 13) Ca^{+2} and Cl^{-1} _____
- 14) Al^{+3} and S^{-2} _____
- 15) Fe^{+2} and S^{-2} _____
- 16) Na^{+1} and NO_3^{-1} _____
- 17) Na^{+1} and SO_4^{-2} _____
- 18) NH_4^{+1} and CO_3^{-2} _____
- 19) Fe^{+3} and CrO_4^{-2} _____
- 20) Al^{+3} and PO_4^{-3} _____

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Introduction to Materials Science for Engineers James F. Shackelford, 2009 For a first course in Materials Sciences and Engineering taught in the departments of materials science mechanical civil and general engineering This text provides balanced current treatment of the full spectrum of engineering materials covering all the physical properties applications and relevant properties associated with engineering materials It explores all of major categories of materials while also offering detailed examinations of a wide range of new materials with high tech applications Publisher's website

Science and Practice of Liquid-liquid Extraction: Process chemistry and extraction operations in the hydrometallurgical, nuclear, pharmaceutical, and food industries John D. Thornton, 1992 This work presents an up to date account of some of the fundamental aspects of liquid liquid extraction technology together with an account of extraction processes in a number of important industries

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