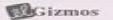
| Activity A: Small atoms | | Get the Gizmo ready: | | |
|----------------------------|--|---|-----------|--|
| | | On the PERIODIC TABLE tab, select H (hydrogen). Select the ELECTRON CONFIGURATION tab. Click Reset. | 1 H 2 | |
| Off | ganization are sho | rons are arranged in orbitals , subshells , and shells . These leaven by the boxes of the Gizmo. Each box represents an orbital d with letters (s , p , d , and t) and the shells are labeled with nur | . The | |
| Q | uestion: How are | electrons arranged in elements with atomic numbers 1 th | rough 10? | |
| 1. | Infer: Based on | its atomic number, how many electrons does a hydrogen atom | have? | |
| 2. | Arrange: The Aufbau principle states that electrons occupy the lowest-energy orbital. Click once in the 1s box to add an electron to the only orbital in the <i>s</i> subshell of the first shell. | | | |
| | Click Check. What is the electron configuration of hydrogen? | | | |
| 3. | Arrange: Click Next element to select helium. Add another electron to the 1s orbital. The arrows represent the spin of the electron. What do you notice about the arrows? | | | |
| | The Pauli exclusion principle states that electrons sharing an orbital have opposite spins. | | | |
| 4. | Check your work | Check your work: Click Check. What is the electron configuration of helium? | | |
| 5. | Arrange: Click Next element and create electron configurations for lithium, beryllium, and boron. Click Check to check your work, and then list each configuration below: | | | |
| | Lithium: | Beryllium: Boron: | | |
| 6. | Arrange: Click Next element to select carbon. Add a second electron to the first 2p orbital. | | | |
| | Click Check. What feedback is given? | | | |
| 7. | Rearrange: Hund's rule states that electrons will occupy an empty orbital when it is available in that subshell. Rearrange the electrons within the 2p subshell and click Check. | | | |
| | Is the configurat | ion correct now? 1s | | |
| | Show the correc | t configuration in the boxes at right: 2s 2p | | |
| | | | | |

(Activity A continued on next page)



Ionic Bonds Explore Learning Answer Key

Lei Shi

Ionic Bonds Explore Learning Answer Key:

The Enigmatic Realm of Ionic Bonds Explore Learning Answer Key: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Ionic Bonds Explore Learning Answer Key** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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