

Ionic & Covalent Compound Naming**Race**

First, identify whether these compounds are ionic or covalent. Then, use the correct formula writing rules to write the correct chemical formulas for each compound.

1)	Compound Name	Type of Compound: Ionic or Covalent	Chemical Formula
	copper (II) chlorite		$\text{Cu}(\text{ClO})$
2)	sodium hydroxide		NaOH
3)	nitrogen dioxide		NO_2
4)	cobalt (III) oxalate		$\text{Co}_2(\text{C}_2\text{O}_4)_3$
5)	ammonium sulfide		$(\text{NH}_4)_2\text{S}$
6)	aluminum cyanide		$\text{Al}(\text{CN})_3$
7)	carbon disulfide		CS_2
8)	tetraphosphorous pentoxide		P_4O_5
9)	potassium permanganate		KMnO_4
10)	manganese (III) chloride		MnCl_3
11)	Compound Name	Type of Compound: Ionic or Covalent	Chemical Formula
	calcium bromate		$\text{Ca}(\text{BrO}_3)_2$
12)	carbon monoxide		CO
13)	potassium oxide		K_2O
14)	antimony tribromide		SbBr_3
15)	zinc phosphate		$\text{Zn}_3(\text{PO}_4)_2$
16)	copper (II) bicarbonate		$\text{Cu}(\text{HCO}_3)_2$
17)	dinitrogen tetroxide		N_2O_4

Ionic And Covalent Nomenclature Ws Answers

Karin Nielsen-Saines



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Table of Contents Ionic And Covalent Nomenclature Ws Answers

1. Understanding the eBook Ionic And Covalent Nomenclature Ws Answers
 - The Rise of Digital Reading Ionic And Covalent Nomenclature Ws Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic And Covalent Nomenclature Ws Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic And Covalent Nomenclature Ws Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic And Covalent Nomenclature Ws Answers
 - Personalized Recommendations
 - Ionic And Covalent Nomenclature Ws Answers User Reviews and Ratings
 - Ionic And Covalent Nomenclature Ws Answers and Bestseller Lists
5. Accessing Ionic And Covalent Nomenclature Ws Answers Free and Paid eBooks
 - Ionic And Covalent Nomenclature Ws Answers Public Domain eBooks
 - Ionic And Covalent Nomenclature Ws Answers eBook Subscription Services
 - Ionic And Covalent Nomenclature Ws Answers Budget-Friendly Options

6. Navigating Ionic And Covalent Nomenclature Ws Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic And Covalent Nomenclature Ws Answers Compatibility with Devices
 - Ionic And Covalent Nomenclature Ws Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic And Covalent Nomenclature Ws Answers
 - Highlighting and Note-Taking Ionic And Covalent Nomenclature Ws Answers
 - Interactive Elements Ionic And Covalent Nomenclature Ws Answers
8. Staying Engaged with Ionic And Covalent Nomenclature Ws Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic And Covalent Nomenclature Ws Answers
9. Balancing eBooks and Physical Books Ionic And Covalent Nomenclature Ws Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic And Covalent Nomenclature Ws Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic And Covalent Nomenclature Ws Answers
 - Setting Reading Goals Ionic And Covalent Nomenclature Ws Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic And Covalent Nomenclature Ws Answers
 - Fact-Checking eBook Content of Ionic And Covalent Nomenclature Ws Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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