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<u>Microeconomics Mcconnell Brue Flynn 19th Edition</u> <u>Summary</u>

DJ Losen

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web choose 1 answer the final temperature is closer to t 1 than to t 2 a the final temperature is closer to t 1 than to t 2 the final temperature is exactly halfway between t 1 and t 2 the final temperature is closer to t 2 than to t 1 c

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water calculate the molar heat of combustion of hexane if the water temperature rises 55~4~c the specific heat capacity of water is 4~184~j~g~c~h~ms~t~h~5~650~grams~h2o~4~184~j~g~c~55~4~c~h~1310~kj calorimetry problems 1 - Feb 17~2022

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