

Study Card: Mass-to-Mass Stoichiometry

Success at solving stoichiometry problems requires ...

1. Understanding of **coefficients** (mole ratios)
Coefficients are used for mole to mole conversions
2. Understanding of **molar mass** (mass/1 mole)
Molar mass allows for a gram \leftrightarrow mole conversion
3. Ability to use the **Factor Label Method** to convert units
Conversion factors are the tool for converting units

Three Step Problem

A \rightarrow B

1. Grams A \rightarrow Moles A
Use Molar Mass of A
2. Moles A \rightarrow Moles B
Use Coefficients
3. Moles B \rightarrow Grams B
Use Molar Mass of B

Method:

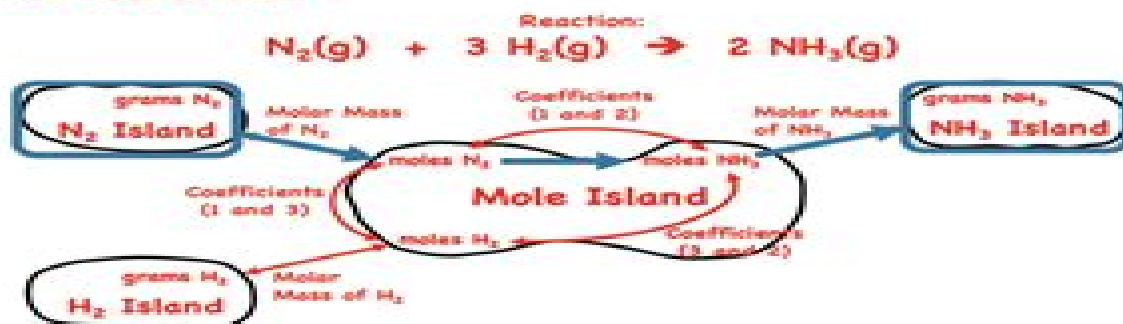
1. **Plan Your Solution**
Use Mole Island to organize your strategy.
2. **Set Up Conversion Factors** (Units; No #s)
Organize your (3) conversion factors to cancel units
3. **Insert Numbers**
Insert molar mass values and coefficients in front of units
4. **Calculate**
Use your calculator to calculate the answer.

Example 1



Determine the mass (in grams) of NH_3 produced from the reaction of 46.2 g of N_2 .

Plan Your Solution



Set Up Conversion Factors (Units, No #s)

$$46.2 \text{ g } \cancel{\text{N}_2} \cdot \frac{1 \text{ mol } \cancel{\text{N}_2}}{28.01 \text{ g } \cancel{\text{N}_2}} \cdot \frac{2 \text{ mol } \text{NH}_3}{1 \text{ mol } \cancel{\text{N}_2}} \cdot \frac{17.03 \text{ g } \text{NH}_3}{1 \text{ mol } \text{NH}_3} = \text{_____ g } \text{NH}_3$$

Insert numbers. Use calculator to solve.

$$46.2 \text{ g } \cancel{\text{N}_2} \cdot \frac{1 \text{ mol } \cancel{\text{N}_2}}{28.01 \text{ g } \cancel{\text{N}_2}} \cdot \frac{2 \text{ mol } \text{NH}_3}{1 \text{ mol } \cancel{\text{N}_2}} \cdot \frac{17.03 \text{ g } \text{NH}_3}{1 \text{ mol } \text{NH}_3} = \boxed{56.2 \text{ g } \text{NH}_3}$$

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Chemistry Tutorial
 \rightarrow Chapter 9
 \rightarrow Lesson 2c



Order is Important

Our observation is that the most frequent student frustrations occur when the early steps are skipped and students rush to their calculator without planning and setting up the solution. The calculator part should be easy. If you are finding it difficult, then you're probably skipping the first two or three steps of the **Method**.

Mass To Mass Stoichiometry Problems Chemfiesta

Ian S. Butler, Arthur E. Grosser



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Bite-Sized Chemistry Calculations John Lambert, 2024-07-10 Bite Sized Chemistry Calculations is a series of books on chemistry calculations aimed at helping students overcome the challenges associated with tackling the various types of calculations encountered in different aspects of chemistry focusing on a few topics at a time to facilitate comprehension. Written by an experienced chemistry educator each book in the series has been tailored to fully meet the needs of students at all levels especially those taking college level general chemistry courses as well as those following various O level curricula worldwide. This part of the series explores the different types of problems and calculations encountered in mass the mole and stoichiometry including the determination of formulae of ionic compounds relative formula masses mass and percent compositions of compounds all aspects of mole calculations empirical and molecular formulae calculations based on chemical equations limiting reagents gas stoichiometry and percent yield. The series is packed with many salient features that are meant to facilitate both teaching and learning. Some of these include helpful explanations many examples alternative ways to solve problems plenty of practice questions complete answers an appendices. With this book you will be well prepared for your exams and boost your performance.

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Stoichiometry Unit Project Luann Marie Decker, 1998

[Chemical Problem-solving by Dimensional Analysis](#) Arnold B. Loebel, 1974

Chemistry Problems David E. Newton, 2001 This edition includes acid base chemistry and thermochemistry. Chemistry Problems is the authoritative resource for practice problems covering all the essentials. Includes Atomic structure Stoichiometry Solutions chemistry and Electrochemistry. Literally thousands of problems in this compendium build proficiency analytical skills and math skills. The text includes a complete answer key and reference to applicable web sites.

Stoichiometry Problems, 2016 You extend your study of stoichiometry to consider more complex problems involving volume molecules and energy.

A Stoichiometry Unit David Callaghan, 2004

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