## Introduction to Periodic Table

| arrest. | arrest to |      | 4.00 | 192     |       |
|---------|-----------|------|------|---------|-------|
| 1       | F-111     | 11.0 | the  | correct | name. |

- a Predicted the properties for elements missing from his periodic table
- b Discovered that each element has a unique atomic number
- C Arranged elements according to the atomic number
- d Arranged elements according to the atomic mass
- Had some elements out of place \_\_\_\_\_\_
- Place all elements in the right place according to properties
- (2) Identify the following elements as metal, nonmetal, and metalloid.
  - Sodium \_\_\_\_\_

6 Chlorine \_\_\_\_

© Silicon \_\_\_\_\_

@ Iron \_\_\_\_\_

Carbon \_\_\_\_\_

- Germanium \_\_\_\_\_
- Write the noble gas configuration for each of the following elements.
  - a Lithium \_\_\_\_\_

© Copper \_\_\_\_\_

d Bromine \_\_\_\_\_

Silver

[] Tellurium

# **Middle School Periodic Table Questions And Answers**

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