

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The particles of the gases are different sizes.)
- At standard atmospheric conditions, all the particles of the same gas have the same kinetic energy.
- At standard atmospheric conditions, the average speed of the particles of a gas is inversely proportional to the square root of its molar mass.

Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of particles in three different phases: solid, liquid, and gas.



Figure 1: Representation of particles in different phases of matter

Model 1 represents the particles in different phases of matter. The particles in the solid phase are closely packed, while the particles in the liquid phase are disordered and the particles in the gaseous phase are widely spaced.

Key Questions

1. Which one of the three representations of particles in different phases of matter is most similar to the particles in the solid phase? Which one is most similar to the particles in the liquid phase? Which one is most similar to the particles in the gaseous phase? Which one is most similar to the particles in the solid phase? Which one is most similar to the particles in the liquid phase? Which one is most similar to the particles in the gaseous phase?

Table 1: Representation of particles in different phases of matter

	Solid	Liquid	Gaseous
Which one of the three representations of particles in different phases of matter is most similar to the particles in the solid phase?	Most similar to the particles in the solid phase. The particles are closely packed and the arrangement is regular.	Most similar to the particles in the liquid phase. The particles are disordered and the arrangement is irregular.	Most similar to the particles in the gaseous phase. The particles are widely spaced and the arrangement is irregular.
Which one of the three representations of particles in different phases of matter is most similar to the particles in the liquid phase?	Most similar to the particles in the solid phase. The particles are closely packed and the arrangement is regular.	Most similar to the particles in the liquid phase. The particles are disordered and the arrangement is irregular.	Most similar to the particles in the gaseous phase. The particles are widely spaced and the arrangement is irregular.
Which one of the three representations of particles in different phases of matter is most similar to the particles in the gaseous phase?	Most similar to the particles in the solid phase. The particles are closely packed and the arrangement is regular.	Most similar to the particles in the liquid phase. The particles are disordered and the arrangement is irregular.	Most similar to the particles in the gaseous phase. The particles are widely spaced and the arrangement is irregular.

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Joseph Oakland Hirschfelder



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Molecules and the Molecular Theory of Matter Allan Douglas Risteen, 1895 **Applications of the Kinetic Theory to Gases, Vapors, Pure Liquids, and the Theory of Solutions** William Pingry Boynton, 1904 **Kinetic Molecular Theory** Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908* An Introduction to Molecular Kinetic Theory Joel Henry Hildebrand, 1963 **Kinetic Theory**, 1923 Volume 1 Kinetic Theory S. G. Brush, 2016-07-04

Kinetic Theory Volume I The Nature of Gases and of Heat deals with kinetic theory and the nature of gases and heat A comprehensive account of the life works and historical environment of a number of scientists such as Robert Boyle and Hermann von Helmholtz is presented This volume is comprised of 11 chapters and begins with an overview of the caloric theory the principle of conservation of energy the virial theorem and atomic magnitudes The discussion then turns to the qualitative atomic theory of the spring of the air proposed by Robert Boyle Isaac Newton's repulsion theory Daniel Bernoulli's theory on the properties and motions of elastic fluids especially air and George Gregory's theory on the existence of fire Subsequent chapters focus on Robert Mayer's theory on the forces of inorganic nature James Joule's theory on matter living force and heat Hermann von Helmholtz's theory on the conservation of force and Rudolf Clausius's theory on the nature of heat James Clerk Maxwell's dynamical theory of gases is also examined This book is written primarily for students and research workers in physics as well as for historians of science **The Molecular Theory of Gases and Liquids** Joseph O. Hirschfelder, Charles F. Curtiss, R. Byron Bird, 1964-01-15 An essential cross disciplinary reference for molecular interactions Molecular Theory of Gases and Liquids offers a rigorous comprehensive treatment of molecular characteristics and behaviors in the gaseous and fluid states A unique cross disciplinary approach provides useful insight for students of chemistry chemical engineering fluid dynamics and a variety of related fields with thorough derivations and in depth explanations throughout Appropriate for graduate students and working scientists alike this book details advanced concepts without sacrificing depth of coverage or technical detail Kinetic Theory J.M Pendlebury, 1985-09-01 The theory of the motions of atoms and molecules in a fluid provides the key to understanding thermodynamics yet many students find difficulty in evaluating and using the basic expressions of kinetic theory This book is designed to lead such students through the fundamental assumptions underlying the theory and to build up a clear physical understanding of the distribution of molecular speeds within a gas The bulk of the text outlines the relationship between molecular motions and the macroscopic properties of gases including the equation of state for a perfect gas and for a van der Waals gas and derives a variety of transport coefficients and thermodynamics relations Molecules and the Molecular Theory of Matter (Classic Reprint) A. D. Risteen, 2017-09-12 Excerpt from Molecules and the Molecular Theory of Matter Throughout this volume I have considered molecules in their physical aspect There are numerous excellent works that discuss the chemical aspect of the molecular theory satisfactorily of which the following may be particularly recommended Meyer's Modern Theories of

Chemistry Rem sen s Theoretical Chemistry Ostwald s Outlines of General Chemistry and Mendeleieff s Principles of Chemistry About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

Kinetic Theory of Bimolecular Chemical Reaction, Diffusive Drag, and Other Processes in a Gas Mixture Richard A. Hord,1967 Molecular Theory of Gases and Liquids Joseph Oakland Hirschfelder,2003-01

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