

PHYSICAL KINETIC MOLECULAR THEORY

Learning Objectives

- to identify the most important differences in behaviour between particles in the three states of matter and to understand the reasons for these differences in behaviour in each state of matter and the processes of the phase transitions theory
- to identify the most important differences between the three states of matter
- to represent quantitatively the three states of matter in a simple way

Module 1: Representation of states in different phases

The following figure shows the representation of the three states of matter in a simple way.



Figure 1: Representation of the three states of matter in a simple way

Modern science has shown that the three states of matter are not so simple. There are many more states of matter than the three states of matter.

Key Concepts

- There are three states of matter: solid, liquid, and gas. The particles in each state are arranged differently. In the solid state, the particles are packed closely together. In the liquid state, the particles are packed less closely together. In the gaseous state, the particles are widely spaced.

Table 1: Representation of the three states of matter

	SOLID STATE	LIQUID STATE	GASEOUS STATE
How the particles are arranged:	The particles are packed closely together. They are in a regular arrangement.	The particles are packed less closely together. They are in a disordered arrangement.	The particles are widely spaced. They are in a disordered arrangement.
How the particles move:	The particles vibrate in place. They do not move from their positions.	The particles move from one position to another. They are in constant motion.	The particles move from one position to another. They are in constant motion.

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M Tight



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Science Citation Index, 1993 Vols for 1964 have guides and journal lists **Molecules and the Molecular Theory of Matter** Allan Douglas Risteen, 1895 **Kinetic Molecular Theory** Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908* **MOLECULES AND THE MOLECULAR THEORY OF MATTER** A. D. RISTEEN, 2018 *The Molecular Theory of Gases and Liquids* Joseph O. Hirschfelder, Charles F. Curtiss, R. Byron Bird, 1964-01-15 An essential cross disciplinary reference for molecular interactions Molecular Theory of Gases and Liquids offers a rigorous comprehensive treatment of molecular characteristics and behaviors in the gaseous and fluid states A unique cross disciplinary approach provides useful insight for students of chemistry chemical engineering fluid dynamics and a variety of related fields with thorough derivations and in depth explanations throughout Appropriate for graduate students and working scientists alike this book details advanced concepts without sacrificing depth of coverage or technical detail [An Introduction to Molecular Kinetic Theory](#) Joel Henry Hildebrand, 1963 [Molecules and the Molecular Theory of Matter \(Classic Reprint\)](#) A. D. Risteen, 2017-09-12 Excerpt from *Molecules and the Molecular Theory of Matter* Throughout this volume I have considered molecules in their physical aspect There are numerous excellent works that discuss the chemical aspect of the molecular theory satisfactorily of which the following may be particularly recommended Meyer s *Modern Theories of Chemistry* Rem sen s *Theoretical Chemistry* Ostwald s *Outlines of General Chemistry* and Mendeleeieff s *Principles of Chemistry* About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works [Molecular Theory of Gases and Liquids](#) Joseph Oakland Hirschfelder, 2003-01
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