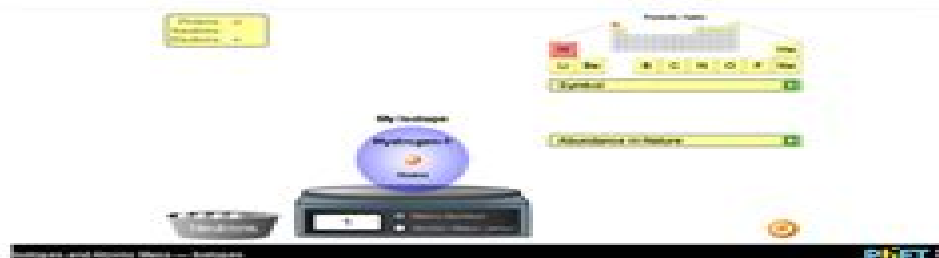


Exploring Isotopes and Atomic Mass - [PhET Simulation](#) [/60]

PART A - Making Isotopes: Open the [Isotopes](#) screen, then explore the simulator for a few minutes to develop your own ideas about "isotopes".



1. What particles determine the mass number? [1]

The neutrons and protons are mainly the particles that determine the mass number and are found in the nucleus of the atom.

2. Why is mass number always a whole number? [1]

Mass number is always a whole number because it's the number of particles in the nucleus, there can't be fractions.

3. What is the approximate mass of one proton? [1]

The approximate mass of one proton is 1.00783 amu

4. What is the approximate mass of one neutron? [1]

The approximate mass of one neutron is 1.00627 amu

5. One isotope of carbon (C) has exactly the same mass number and atomic mass since it was used as the definition of the atomic mass unit (amu). Which isotope is it and what is its atomic mass? [2]

The isotope is Carbon-12 and its atomic mass is 12

6. Based on what you learned from the simulator, what is the definition of an isotope? *In your definition, include the terms: mass number, atomic number, number of protons (p^+), neutrons (n^0) and electrons (e^-).* [3]

Isotopes And Atomic Mass Phet Lab Answers

Vladimir Peter Derenchuk



Isotopes And Atomic Mass Phet Lab Answers:

Government Reports Announcements & Index, 1992 *Table of Isotopes* Richard B. Firestone, 1999-08-03 Nuclear structure and decay data for thousands of isotopes and isomers a new update of the definitive reference The 1999 Table of Isotopes booklet set features Nuclear structure and radioactive decay data for approximately 3 700 isotopes and isomers an increase of more than 100 since the 1998 Update Up to date mass chain information with more than 10% revised material The latest versions as of December 1998 of the Evaluated Nuclear Structure Data File ENSDF and Nuclear Science Reference NSR file Additional data from several evaluation sources including The Table of Superdeformed Nuclear Bands and Fission Isomers Updated isotope summary table as well as energy ordered gamma ray and alpha particle tables now included on the software available from the book s website at www.wiley-vch.de/books/info/0_471_35633_6 Updated appendices for elemental data nuclear charts and gamma ray energy standards Adoption of the 1997 IUPAC recommended heavy element names More than 25 000 references Convenient links to additional atomic mass nuclear astrophysics rates spontaneous fission thermal neutron capture and more Plus the Isotope Explorer 2 22 software lets you search the entire database by level scheme drawings annotated tables data plots nuclear structure charts and keywords as well as download the latest data directly from the Table of Isotopes Web site The 1999 Update booklet features a new comprehensive isotope nuclear structure table In addition it provides clear step by step instructions on navigating the Table of Isotopes electronic content and accessing its Web site System Requirements PC Macintosh r or UNIX r systems with double speed and sufficient RAM to run Adobe TM Acrobat r see Adobe Acrobat Reader information in the book for specific system requirements Included Adobe Acrobat Reader r 3 02 for Windows r 95 98 and NT and Linux 1 2 13 or higher Acrobat Reader 3 01 for Windows 3 1 Isotope Explorer 2 22 for Windows 95 98 and NT **The Masses of Heavy Isotopes** Martin O. Stern, 1948
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 - Chapter 2: Essential Elements of Isotopes And Atomic Mass Phet Lab Answers
 - Chapter 3: Isotopes And Atomic Mass Phet Lab Answers in Everyday Life
 - Chapter 4: Isotopes And Atomic Mass Phet Lab Answers in Specific Contexts
 - Chapter 5: Conclusion
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