

## 7. Charles's Law

### Background

#### Testing Charles's Law

Gases are different from liquids and solids in that they have neither a fixed volume nor a fixed shape. The behavior of gases depends on four variables: *pressure* ( $P$ ), *volume* ( $V$ ), *temperature* ( $T$ ), and *amount* (number of moles,  $n$ ). Three key laws express the relationship between these variables: *Boyle's law*, *Charles's law*, and *Avogadro's law*. These laws express the effect of one variable on another, while the other two variables remain constant. Because volume is relatively easy to measure compared to the other variables, the gas laws are expressed as the effect of one of the other variables on gas volume. Charles's law describes the relationship between the temperature of a gas and its volume.

#### Charles's Law Describes the Effect of Temperature on Volume

Jacques Charles was a physics professor at Sorbonne University in France as well as a hot air balloon enthusiast. Hot air balloons rise when the gas inside is heated, which causes the gas to expand in volume and become less dense. Charles decided to investigate the relationship between gas volume and temperature at constant pressure in his laboratory. Charles's law, proposed in the late eighteenth century, states that *at constant pressure the volume occupied by a fixed amount of gas is directly proportional to its temperature*. In other words, as the temperature increases, the volume increases proportionally and vice versa. This is expressed mathematically using the following equations.

# Lab Manual Charles Law

**Robert C. Smoot, Smoot, Richard G.  
Smith, Jack Price**



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**A Laboratory Manual to Accompany Godfrey's Elementary Chemistry** Hollis Godfrey, 1910      **A Laboratory Manual of General Chemistry** James Henri Walton, Francis Craig Krauskopf, 1920      **The Essential Lab Manual** Karen Timberlake, 2002-06-24 Drawing from the successful main Laboratory Manual the Essential Laboratory Manual includes twenty one experiments which have been revised and updated Suitable for a one or two term lab course      **Laboratory Manual and Principles of Chemistry for Beginners** George Mann Richardson, 1894      **A Laboratory Manual of Physics and Applied Electricity: Senior courses and outlines of advanced work** Edward Leamington Nichols, 1896

A Laboratory Manual of General Chemistry Leon Burr Richardson, Andrew Jackson Scarlett, 1929      Laboratory Manual of Inorganic Chemistry for Colleges Lyman Churchill Newell, 1916      **A Laboratory Manual of Experiments in Physics** Leonard Rose Ingersoll, 1925      **A Laboratory Manual of Qualitative Chemical Analysis** Hugo Herman Schaefer, Andrew Richard Bliss, 1928      **Laboratory Manual for College Chemistry** William E. Bull, 1980      *Laboratory Manual: Dynamic Physics* Ernest Oscar Bower, Edward P. Robinson, 1943      A Laboratory manual of elemental qualitative chemical analysis Andrew Richard Bliss, 1918      **Laboratory Manual for College Chemistry** William T. Smith, Jesse Hermon Wood, 1966

**Laboratory Manual of the Technic of Basal Metabolic Rate Determinations** Walter Meredith Boothby, Irene Sandiford, 1920      Laboratory Manual of General Inorganic Chemistry Mayce Cannon Sneed, Raymond Eller Kirk, 1926

Laboratory Manual of Inorganic Chemistry and Elementary Qualitative Analysis C. C. Hedges, 1937      *Laboratory Manual to Accompany Chemistry, [by] Stanley R. Radel, Marjorie H. Navidi* Arthur D. Baker, 1990      **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998      **Laboratory Manual for General College Chemistry** Jesse Hermon Wood, William T. Smith, 1957      Clinical Chemistry: Laboratory Manual and Methods Joseph H. Boutwell, 1961

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the old republic era also referred to as the sith era takes place from 25 053 bby the founding of the republic until 1000 bby the seventh battle of ruusan and the end of the new sith wars tilotny throws a shape the empire strikes back monthly 154

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knights of the old republic ii was developed by obsidian entertainment and published by lucasarts it is the sequel to bioware's knights of the old republic and it uses the same odyssey engine as the original game

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web jan 5 2022 the chevy s10 starter wiring diagram is an essential reference for anyone who needs to identify the different components of a chevy s10 starter and understand how they are connected to each other this diagram is especially helpful when attempting to diagnose electrical problems

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web locate the air filter housing and remove the cover remove the old air filter and clean the housing if necessary install the new air filter and replace the cover check the filter periodically and replace as needed how do i replace the alternator in my chevrolet s 10