

## Moles and Chemical Formulas

5

## A. Finding the Simplest Formula

1. Mass of empty crucible + cover	<u>43.58</u>	g
2. Initial appearance of the magnesium	<u>Silver shiny solid</u>	
3. Mass of crucible + cover + magnesium	<u>43.81</u>	g
4. Mass of crucible + cover + oxide product	<u>43.95</u>	g

## Calculations

5. Mass of magnesium	<u><math>43.81 - 43.58 = .23</math></u>	g
6. Mass of magnesium compound	<u><math>43.95 - 43.58 = .37</math></u>	g
7. Mass of oxygen in the product	<u><math>.37 - .23 = .14</math></u>	g
8. Moles of Mg	<u>.00946</u>	mole

(Show calculations.)

$$\frac{.23 \text{ g}}{24.31 \text{ g/mol}} = .00946112 \text{ or } 9.46 \times 10^{-3} \text{ mol}$$

9. Moles of O	<u>.00875</u>	mole
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(Show calculations.)

$$\frac{.14 \text{ g}}{15.99 \text{ g/mol}} = .00875547 \text{ or } 8.76 \times 10^{-3} \text{ mol}$$

10. Which number of moles (Mg or O) is smaller Oxygen

     moles of Mg = .0095 moles of Mg (rounded to a whole number)

    

     moles O = .0088 moles of O (rounded to a whole number)

    

Formula: Mg      O      subscripts

# Lab Report On Moles And Chemical Formulas

**Roman Wölfel**



## **Lab Report On Moles And Chemical Formulas:**

Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helps each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study. Features: Each suggested weekly schedule has five easy-to-manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated, and three-hole punched materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates. There is flexibility in scheduling. Adapt the days to your school schedule. Workflow: Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade. About the Author: DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

**Exploring General, Organic, & Biochemistry in the Laboratory** William G. O'Neal, 2017-02-01 This full-color comprehensive affordable manual is appropriate for two-semester introductory chemistry courses. It is loaded with clearly written exercises, critical thinking questions, and full-color illustrations and photographs providing ample visual support for experiment set-up, technique, and results.

**General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text, this guide is designed to promote active learning through a variety of exercises with answers and mastery exams. The guide also contains complete solutions to odd-numbered problems.

**Chemical Investigations** Nancy Konigsberg Kerner, 1986 *Chemistry in the Laboratory* James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written, class-tested manual has long given students

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with a wealth of knowledge and insight into Dr Herron s experiences in teaching and learning chemistry Using specific examples from chemistry to illustrate principles of learning the volume applies cognitive science to teaching chemistry and explores such topics as how individuals learn teaching problem solving concept learning language roles and task involvement Includes learning exercises to help educators decide how they should teach

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