

Can You Find the Needle in the Hay?

In this experiment, feeding trials (FHTC) involve both hydrocarbon and (H-C) as growth medium; a substrate, carbon dioxide, and water. You will determine the mass of substrate used and mass of product produced using both the lab data and stoichiometry and compare this to:

Stoichiometric

mass of reactants used feeding trials	moles
mass of substrate used	moles
mass of product used	moles
mass of reactants used stoichiometric	moles
mass of substrate used	moles
mass of product used	moles

Calculate Questions:

- Calculate the number of moles of feeding trials used in the lab. (100)

$$\frac{\text{mass of substrate used}}{\text{molar mass of substrate}}$$
- Calculate the number of moles of product substrate produced in the lab.

$$\frac{\text{mass of product used}}{\text{molar mass of product}}$$
- What is the experimental ratio of feeding trials to product substrate (FHTC)?

$$\frac{\text{moles of substrate used}}{\text{moles of product used}}$$
- What is stoichiometric ratio for the reaction that took place in the experiment?

$$\frac{\text{mass of substrate used}}{\text{molar mass of substrate}} = \frac{\text{mass of product used}}{\text{molar mass of product}}$$
- According to the stoichiometric ratio, what is the theoretical ratio of feeding trials to product substrate?

$$\frac{\text{mass of substrate used}}{\text{molar mass of substrate}} = \frac{\text{mass of product used}}{\text{molar mass of product}}$$
- What the experimental ratio is equal to the ratio of the theoretical ratio, what is the ratio of the experimental ratio to the theoretical ratio?

$$\frac{\text{mass of substrate used}}{\text{molar mass of substrate}} = \frac{\text{mass of product used}}{\text{molar mass of product}}$$

Lab Episode 801 Datasheet Answers

Patrick Vollmar



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