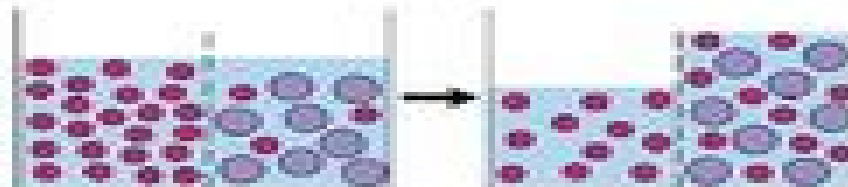


## Gummy Bear Osmosis Lab

**Background:** Molecules are in constant motion. As they move around they tend to move towards equilibrium. This means that the molecules want to even out their concentrations. So, molecules tend to move from areas of high concentration to areas of lower concentration, until the concentrations are even. This process requires no energy and is called **DIFFUSION**. Molecules continue to move even when equilibrium is reached, but they move equally in both directions. There is no net movement in any one direction. This is called dynamic equilibrium.



Osmosis is similar to simple diffusion, but it is specific to water molecules moving through a selectively permeable membrane. Water molecules follow the rules of diffusion, so they move from where there are more water molecules to where there are less. When another substance such as starch is added into a solution of water, the concentration of free water molecules decreases, so water molecules on the other side of the membrane where there is pure water will flow across the membrane into the starch solution where there are less water molecules.

Osmosis and diffusion are both forms of passive transport because they require no energy. Gummy Bears are made up of gelatin, starch and sugar. The gelatin acts like a semi-permeable membrane. Inside the membrane the gummy bear is full of starch and sugar (mostly sugar).



# Lab Report Gummy Bear Experiment Osmosis

**Léo Pomar**



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