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$\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2}\text{O}_2(\text{g})$ $\Delta h = 71.2 \text{ kJ}$

$\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2}\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3\text{H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$

1 $\text{C}_2\text{H}_6(\text{g})$

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