

$$\text{LR.10} \quad 1.5 \text{ L} \cdot \frac{10^3 \text{ mL}}{1 \text{ L}} = 1.5 \times 10^3 \text{ mL}$$

$$1.5 \text{ L} \cdot \frac{10^3 \text{ mL}}{1 \text{ L}} \cdot \frac{1 \text{ cm}^3}{1 \text{ mL}} = 1.5 \times 10^3 \text{ cm}^3$$

$$1.5 \times 10^3 \text{ cm}^3 \cdot \left( \frac{1 \text{ dm}}{10 \text{ cm}} \right)^3 = 1.5 \text{ dm}^3$$

$$\text{LR.11} \quad 2.52 \text{ kg} \cdot \frac{10^3 \text{ g}}{1 \text{ kg}} = 2.52 \times 10^3 \text{ g}$$

$$\text{LR.12} \quad 2.265 \text{ g} \cdot \frac{1 \text{ kg}}{10^3 \text{ g}} = 2.265 \times 10^{-3} \text{ kg}$$

$$2.265 \text{ g} \cdot \frac{10^3 \text{ mg}}{1 \text{ g}} = 2.265 \times 10^3 \text{ mg}$$

$$\text{LR.13} \quad 500. \text{ mL} \cdot \frac{1 \text{ cm}^3}{1 \text{ mL}} \cdot \frac{1.11 \text{ g}}{1 \text{ cm}^3} = 555 \text{ g}$$

$$\text{LR.14} \quad 2.365 \text{ g} \cdot \frac{1 \text{ cm}^3}{10.5 \text{ g}} = 0.225 \text{ cm}^3$$

$$\text{LR.15} \quad \text{density} = \frac{2.361 \text{ g}}{(2.35 \text{ cm} \times 1.34 \text{ cm} \times 0.105 \text{ cm})} = 7.14 \text{ g/cm}^3$$

The metal is (c) Zinc

$$\text{LR.16} \quad 600 \text{ g H}_2\text{O} \cdot \frac{1 \text{ cm}^3}{0.995 \text{ g}} = 600 \text{ cm}^3$$

$$600 \text{ g Pb} \cdot \frac{1 \text{ cm}^3}{11.35 \text{ g}} = 50 \text{ cm}^3$$

Given equal masses the less dense substance (water) will have a greater volume.

$$\text{LR.17} \quad 1200 \text{ Cal} \cdot \frac{10^3 \text{ cal}}{1 \text{ Cal}} \cdot \frac{4.184 \text{ J}}{1 \text{ cal}} = 5.0 \times 10^6 \text{ J}$$

$$\text{LR.18} \quad 1670 \text{ kJ} \cdot \frac{10^3 \text{ J}}{1 \text{ kJ}} \cdot \frac{1 \text{ cal}}{4.184 \text{ J}} \cdot \frac{1 \text{ Cal}}{10^3 \text{ cal}} = 399 \text{ Cal}$$

$$\text{LR.19} \quad 170 \text{ kcal} \cdot \frac{10^3 \text{ cal}}{1 \text{ kcal}} \cdot \frac{4.184 \text{ J}}{1 \text{ cal}} \cdot \frac{1 \text{ kJ}}{10^3 \text{ J}} = 710 \text{ kJ}$$

The food product with 170 kcal per serving has a greater energy content per serving.

# John Townsend Solution Manual

**Michael Rajnik**



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