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[\$\text{hno}_3 + \text{mg} + 2 \text{c}_2\text{h}_3\text{o}_2 \rightarrow 2 \text{hc}_2\text{h}_3\text{o}_2 + \text{mg} + \text{no}_3 + 2\$ chemical equation](#) - Aug 11 2023

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web how to balance $\text{HNO}_3 + \text{Mg} \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2$ word equation nitric acid magnesium magnesium nitrate hydrogen gas type of chemical reaction for this reaction we have a single displacement reaction balancing strategies in this single displacement reaction the mg replaces the h in the HNO_3

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web nov 20 2020 in this video we ll balance the equation $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2 + \text{H}_2\text{O}$ and provide the correct coefficients for each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2 + \text{H}_2\text{O}$ you ll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{Mg}(\text{OH})_2(\text{aq}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}$ total ionic equation

$\text{Mg}^{2+} + 2\text{OH}^{-} + 1\text{H}^{+} + 2\text{NO}_3^{-} + 1\text{H}_2\text{O} + \text{Mg}^{2+} + 2\text{NO}_3^{-}$ net ionic equation
 $2\text{OH}^{-} + 1\text{H}^{+} + 1\text{H}_2\text{O} + \text{Sulfuric acid} + \text{Lithium hydroxide}$
 balanced total equation
 $\text{H}_2\text{SO}_4 + 2\text{LiOH} \rightarrow \text{Li}_2\text{SO}_4 + 2\text{H}_2\text{O}$