

Metals in Aqueous Solutions Simulations

How to use it:

Some elements are more reactive than others, meaning that they can more easily be broken apart into their ions than others. These are listed in their place respectively in "less" reactive. Other elements are less reactive and more likely to be broken apart into ions. This simulation will allow you to test various combinations of elements and compounds, as well as reactions of elements, to see how they react together and how they react with water or other substances. Some reactions will produce a precipitate, some will produce a gas, and some will produce a solution. Some reactions will produce a precipitate, some will produce a gas, and some will produce a solution. Some reactions will produce a precipitate, some will produce a gas, and some will produce a solution.

Objectives:

Use the results of the simulation to determine the relative reactivity of the elements and compounds.

Procedure:

For this, the simulation is: <https://www.ck12.org/chemistry/transition-metal-complexes/transition-metal-complexes-simulation/> and complete parts of the PCOP, including (a) and (b) and the Observation section, recording the appropriate information and answering the questions below.

Metals In Aqueous Solutions Answers

M Walker



Metals In Aqueous Solutions Answers:

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