

Can You Find the Side Sides?

In this experiment, boiling acids (H₂SO₄/H₂O) react with hydrocarbons and (HCl) to produce various alkenes, various alcohols, and water. You will determine the ratios of reactants used and ratios of products produced using both the lab data and stoichiometry and compare this to:

Side Table:

Ratio of reactants used/boiling acids	Ratio
Ratio of alky/alkene ratio	Ratio
Ratio of alky/alkene ratio	Ratio
Ratio of alky/alkene ratio/alkene/alcohol	Ratio
Ratio of alky/alkene ratio	Ratio
Ratio of alky/alkene ratio	Ratio

Calculations/Questions:

- Calculate the number of ratios of boiling acids used in this lab. (100%

$$\frac{\text{Boiling acids}}{\text{Boiling acids}} = \frac{\text{Boiling acids}}{\text{Boiling acids}}$$
- Calculate the number of ratios of products produced in this lab.

$$\frac{\text{Alky/alkene}}{\text{Alky/alkene}} = \frac{\text{Alky/alkene}}{\text{Alky/alkene}}$$
- What is the experimental ratio ratio of boiling acids (H₂SO₄/H₂O) to alkene/alcohol (H₂O)?

$$\frac{\text{Boiling acids}}{\text{Alky/alkene}}$$
- What is the experimental ratio for the number of alky/alkene in the experiment?

$$\frac{\text{Boiling acids}}{\text{Alky/alkene}} = \frac{\text{Alky/alkene}}{\text{Alky/alkene}}$$
- According to the experimental ratio, what is the theoretical ratio ratio of boiling acids to alkene/alcohol?

$$\frac{\text{Alky/alkene}}{\text{Alky/alkene}}$$
- What the experimental ratio ratio used by this ratio in the theoretical ratio ratio? Or not, give some possible reasons.

$$\frac{\text{Boiling acids}}{\text{Alky/alkene}} = \frac{\text{Alky/alkene}}{\text{Alky/alkene}}$$

Lab Episode 801 Datasheet Answers

Didier Musso



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