# GRAPHING TRENDS OF THE PERIODIC TABLE

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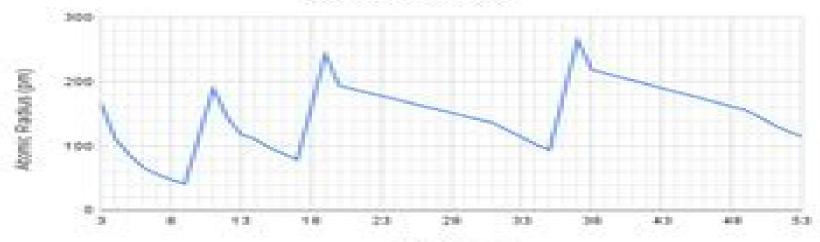
NAME		pare.
PERIOD.	Graphing Periodic Table Trends	

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Asomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
- 1.00 num - 3 - 1	567	Sodium - 11	190	Protessium - 19	243	Plubidium - 37	295
Blancy Blazami - 41	112	Magnessum - 12	145	California 20	104	Strombum - 28	27100
Biomom - S-	87	Alduminum - 13	1118	Gallium - 31	136	indium : 49	156
Carbon - 6	67	Sittope - 14	211	Clermanium - 32	125	Tm - 50	7405
Nitregen - F	866	Phosphorus i 15	94	Acuerisc - 30	2004	Actionomy - 51	153
Chargen - B	40.	Sutter - 16	888	Sieterium - 34	100	Tellunum - 52	1.23
Fluorine - 9	42	Chlorine - 17	20 PM 15	Binomine : 35	1946	todane - 53	1100

Tomication Emergy (k./mol)							
1.8houm - 3	6.250	Soldham - 11	400	Processoum - 19	479	Plubabum - 37	44000
Benyflum - 4	9000	Magnesture + 12	730	Calcium - 20	590	Sportfurn - 38	10000
		The second second second second		The second second second			

# Atomic Radius



Adomic Number

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**R Pring** 

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